

UC DAVIS 2000-054-6

**IN THE UNITED STATES PATENT AND TRADEMARK OFFICE****APPLICATION FOR PATENT****THERAPEUTIC PLATELETS AND METHODS**

Inventors: John H. Crowe  
Fern Tablin  
Willem F. Wolkers  
Naomi J. Walker

**Related Patent Applications**

This is a continuation-in-part patent application of co-pending patent application Serial No. 10/052,162, filed January 16, 2002. Patent application Serial No. 10/052,162 is a continuation-in-part patent application of co-pending patent application Serial No. 09/927,760, filed August 9, 2001. Patent application Serial No. 09/927,760 is a continuation-in-part patent application of co-pending patent application Serial No. 09/828,627, filed April 5, 2001. Patent application Serial No. 09/828,627 is a continuation patent application of patent application Serial No. 09/501,773, filed February 10, 2000. Benefit of all earlier filing dates is claimed with respect to all common subject matter.

### Field of the Invention

Embodiments of the present invention generally broadly relate to living mammalian cells. More specifically, embodiments of the present invention generally provide for the preservation and survival of human cells, especially eukaryotic cells and erythrocytic cells.

Embodiments of the present invention also generally broadly relate to the therapeutic uses of blood platelets, eukaryotic cells, and erythrocytic cells; and more particularly to manipulations or modifications of platelets, eukaryotic cells, and erythrocytic cells, such as in preparing freeze-dried compositions that can be rehydrated at the time of application. When freeze-dried platelets are rehydrated, they have a normal response to thrombin and other agonists with respect to that of fresh platelets. When eukaryotic cells and erythrocytic cells are rehydrated, they are immediately restored to viability.

The inventive compositions and methods for embodiments of the present invention are useful in many applications, such as in medicine, pharmaceuticals, biotechnology, and agriculture, and including transfusion therapy, as hemostasis aids and for drug delivery.

### Statement Regarding Federal Sponsored Research and Development

Embodiments of this invention were made with Government support under Grant No. N66001-00-C-8048, awarded by the Department of Defense Advanced Research Projects Agency (DARPA). Further embodiments of this invention were made with Government support under Grant Nos. HL57810 and HL61204, awarded by the National Institutes of Health. The Government has certain rights to embodiments of this invention.

### Background of the Invention

Blood transfusion centers are under considerable pressure to produce platelet concentrates for transfusion. The enormous quest for platelets necessitates storage of this blood component, since platelets are important contributors to hemostasis. Platelets are generally oval to spherical in shape and have a diameter of 2-4  $\mu\text{m}$ . Today platelet rich plasma concentrates are stored in blood bags at 22° C; however, the shelf life under these conditions is limited to five days. The rapid loss of platelet function during storage and risk of bacterial contamination complicates distribution and availability of platelet concentrates. Platelets tend to become activated at low temperatures. When activated they are substantially useless for an application such as transfusion therapy. Therefore the development of preservation methods that will increase platelet lifespan is desirable.

Several techniques for preservation of platelets have been developed over the past few decades. Cryopreservation of platelets using various agents, such as glycerol (Valeri et al., *Blood*, 43, 131-136, 1974) or dimethyl sulfoxide, "DMSO" (Bock et al., *Transfusion*, 35, 921-924, 1995), as the cryoprotectant have been done with some success. The best results have been obtained with DMSO. However, a considerable fraction of these cells are partly lysed after thawing and have the shape of a balloon. These balloon cells are not responsive to various agonists, so that overall responsiveness of frozen thawed platelets to various agonists is reduced to less than 35% compared with fresh platelets. The shelf life of cryopreserved DMSO platelets at -80°C is reported to be one year, but requires extensive washing and processing to remove cryoprotective agents, and even then

the final product has a severe reduction in ability to form a clot.

Attempts to dry platelets by lyophilization have been described with paraformaldehyde fixed platelets (Read et al., *Proc. Natl. Acad. Sci. USA*, 92, 397401, 1995). U.S. Patent No. 5,902,608, issued May 11, 1999, inventors Read et al. describe and claim a surgical aid comprising a substrate on which fixed, dried blood platelets are carried. These dried blood platelets are fixed by contacting the platelets to a fixative such as formaldehyde, paraformaldehyde, glutaraldehyde, or permanganate. Proper functioning of lyophilized platelets that have been fixed by such fixative agents in hemostasis is questionable.

Spargo et al., U.S. Patent No. 5,736,313, issued April 7, 1998, have described a method in which platelets are loaded overnight with an agent, preferably glucose, and subsequently lyophilized. The platelets are preincubated in a preincubation buffer and then are loaded with carbohydrate, preferably glucose, having a concentration in the range of about 100 mM to about 1.5 M. The incubation is taught to be conducted at about 10°C to about 37°C, most preferably about 25° C.

U.S. Patent No. 5,827,741, Beattie et al., issued October 27, 1998, discloses cryoprotectants for human platelets, such as dimethylsulfoxide and trehalose. The platelets may be suspended, for example, in a solution containing a cryoprotectant at a temperature of about 22°C and then cooled to below 15°C. This incorporates some cryoprotectant into the cells. Trehalose is a disaccharide found at high concentrations in a wide variety of organisms that are capable of surviving almost complete dehydration (Crowe et al., *Anhydrobiosis. Annul. Rev. Physiol*, 54, 579-599, 1992). Trehalose has been shown to stabilize certain cells during freezing and drying (Leslie et al.,

*Biochim. Biophys. Acta*, 1192, 7-13, 1994; Beattie et al., *Diabetes*, 46, 519-523, 1997).

Other workers have sought to load platelets with trehalose through use of electroporation before drying under vacuum. However, electroporation is very damaging to the cell membranes and is believed to activate the platelets. Activated platelets have dubious clinical value.

Platelets have also been suggested for drug delivery applications in the treatment of various diseases, as is discussed by U.S. Patent No. 5, 759,542, issued June 2, 1998, inventor Gurewich. This patent discloses the preparation of a complex formed from a fusion drug including an A-chain of a urokinase-type plasminogen activator that is bound to an outer membrane of a platelet.

Accordingly, a need exists for the effective and efficient preservation of platelets such that they maintain, or preserve, their biological properties, particularly their response to platelet agonists such as thrombin, and which can be practiced on a large, commercially feasible scale. Further, it would also be useful to expand the types of present vehicles that are useful for encapsulating drugs and used for drug delivery to targeted sites. Accordingly further, a need also exists for the effective and efficient preservation of eukaryotic cells and erythrocytic cells, such that these cells respectively maintain their biological properties and may readily become viable after storage.

### Summary of Embodiments of the Invention

In one aspect of the present invention, a dehydrated composition is provided comprising freeze-dried platelets that are effectively loaded with trehalose to preserve biological properties during freeze-drying and rehydration. These platelets are rehydratable so as to have a normal response to at least one agonist, such as thrombin. For example, substantially all freeze-dried platelets of the invention when rehydrated and mixed with thrombin (1 U/ml) form a clot within three minutes at 37° C. The dehydrated composition can include one or more other agents, such as antibiotics, antifungals, growth factors, or the like, depending upon the desired therapeutic application.

Embodiments of the present invention provide a process for preparing a dehydrated composition comprising disposing platelets in an oligosaccharide solution for loading an oligosaccharide from the oligosaccharide solution into the platelets, preventing a decrease in a loading efficiency gradient in the loading of the oligosaccharide into the platelets, and lyophilizing the platelets. The preventing a decrease in a loading efficiency gradient in the loading of the oligosaccharide into the platelets may comprise maintaining a concentration of the oligosaccharide in the oligosaccharide solution below about 50 mM. The preventing a decrease in a loading efficiency gradient in the loading of the oligosaccharide into the platelets may also comprise maintaining a positive gradient of loading efficiency (%) to concentration (mM) of the oligosaccharide in the oligosaccharide solution.

Embodiments of the present invention also provide a process for preparing a dehydrated composition comprising disposing platelets in an oligosaccharide solution for loading an oligosaccharide from the oligosaccharide solution into the

platelets, preventing a decrease in a loading gradient in the loading of the oligosaccharide into the platelets, and lyophilizing the platelets. The preventing a decrease in a loading gradient in the loading of the oligosaccharide into the platelets may comprise maintaining a concentration of the oligosaccharide in the oligosaccharide solution below about 50 mM. The preventing a decrease in a loading gradient in the loading of the oligosaccharide into the platelets may also comprise maintaining a positive gradient of concentration of oligosaccharide loaded into the platelets to concentration of the oligosaccharide in the oligosaccharide solution.

In another aspect of the present invention, a hemostasis aid is provided where the above described freeze-dried platelets are carried on or by a biocompatible surface. A further component of the hemostasis aid may be a therapeutic agent, such as an antibiotic, an antifungal, or a growth factor. The biocompatible surface may be a bandage or a thrombic surface, such as freeze-dried collagen. Such a hemostasis aid can be rehydrated just before the time of application, such as by hydrating the surface on or by which the platelets are carried, or, in case of an emergency, the dry hemostasis treatment aid could be applied directly to the wound or burn and hydrated *in situ*.

Methods of making and using inventive embodiments are also described. One such method is a process of preparing a dehydrated composition comprising providing a source of platelets, effectively loading the platelets with trehalose to preserve biological properties, cooling the trehalose loaded platelets to below their freezing point, and lyophilizing the cooled platelets. The trehalose loading includes incubating the platelets at a temperature from greater than about 25°C to less

than about 40°C with a trehalose solution having up to about 50 mm trehalose therein. The process of using such a dehydrated composition further may comprise rehydrating the platelets. The rehydration preferably includes a prehydration step wherein the freeze-dried platelets are exposed to warm, saturated air for a time sufficient to bring the water content of the freeze-dried platelets to between about 20 weight percent to about 35 weight percent.

In yet another aspect of the present invention, a drug delivery composition is provided comprising platelets having a homogeneously distributed concentration of a therapeutic agent therein. The drug delivery composition is particularly useful for targeting the encapsulated drug to platelet-mediated sites.

Practice of the present invention permits the manipulation or modification of platelets while maintaining, or preserving, biological properties, such as a response to thrombin. Further, use of the method to preserve platelets can be practiced on a large, commercially feasible scale, and avoids platelet activation. The inventive freeze-dried platelets, and hemostasis aids including the freeze-dried platelets, are substantially shelf stable at ambient temperatures when packaged in moisture barrier materials.

Embodiments of the present invention also provide a process for preserving and/or increasing the survival of dehydrated eukaryotic cells after storage comprising providing eukaryotic cells from a mammalian species (e.g., a human); loading the eukaryotic cells with a preservative (e.g., an oligosaccharide, such as trehalose); dehydrating the eukaryotic cells while maintaining a residual water content in the eukaryotic cells greater than about 0.15 (e.g., from about 0.20 to about 0.75) gram of water per gram of dry weight eukaryotic cells to



increase eukaryotic cell survival, preferably to greater than about 80%, upon rehydrating after storage; storing the dehydrated eukaryotic cells having the residual water content greater than about 0.15 gram of water per gram of dry weight eukaryotic cells; and rehydrating the stored dehydrated eukaryotic cells with the stored dehydrated eukaryotic cells having an increase in survival following dehydration and storage. In a preferred embodiment, more than about 80% of the stored dehydrated cells survive the dehydration and storage.

Embodiments of the present invention further provide a process of preparing loaded eukaryotic cells comprising providing eukaryotic cells selected from a mammalian species; and loading (e.g., with an oligosaccharide solution and/or with or without a fixative) an oligosaccharide (e.g., trehalose) into the eukaryotic cells at a temperature greater than about 25°C (e.g., greater than about 25°C but less than about 50°C, such as from about 30°C to less than about 50°C, or from about 30°C to about 40°C) to produce loaded eukaryotic cells. The loading comprises taking up external oligosaccharide via fluid phase endocytosis from an oligosaccharide solution at the temperature greater than about 25°C. The loading further comprises incubating the eukaryotic cells at the temperature greater than about 25°C with the oligosaccharide solution. For these embodiments of the present invention, the eukaryotic cells are preferably human eukaryotic cells, such as, by way of example only, eukaryotic cells selected from the group of eukaryotic cells consisting of mesenchymal stem cells and epithelial 293H cells.

Embodiments of the present invention also further provide a solution for loading eukaryotic cells comprising eukaryotic cells selected from a mammalian species; and an oligosaccharide

solution containing the eukaryotic cells and a temperature greater than about 25°C for loading oligosaccharide from the oligosaccharide solution into the eukaryotic cells. External oligosaccharide is taken up via fluid phase endocytosis from the oligosaccharide solution at a temperature ranging from about 30°C to less than about 42°C. A eukaryotic cell composition is also provided as broadly comprising eukaryotic cells loaded internally with an oligosaccharide, preferably trehalose, from an oligosaccharide solution at a temperature greater than about 25°C.

Embodiments of the present invention yet also further provide a generally dehydrated composition comprising freeze-dried eukaryotic cells selected from a mammalian species (e.g., a human) and being effectively loaded internally (e.g., incubating the eukaryotic cells at a temperature from about 30°C to less than about 50°C so as to uptake external trehalose via fluid phase endocytosis) with at least about 10 mM trehalose therein to preserve biological properties during freeze-drying and rehydration. The amount of trehalose loaded inside the freeze-dried eukaryotic cells is preferably from about 10 mM to about 50 mM. The freeze-dried eukaryotic cells comprise at least about 0.15 (e.g., from about 0.20 to about 0.75) gram of residual water per gram of dry weight eukaryotic cells to increase eukaryotic cell survival upon rehydrating.

Aspects of embodiments of the present invention also include a process for preparing a dehydrated composition. The process comprises providing eukaryotic cells selected from a mammalian species (e.g., a human); loading internally the eukaryotic cells with from about 10 mM to about 50 mM of an oligosaccharide (e.g., trehalose) therein to preserve biological properties. The loading includes incubating the eukaryotic cells

at a temperature from about 30°C to less than about 50°C, preferably from about 30°C to about 40°C, more preferably from about 34°C to about 37°C, with an oligosaccharide solution having up to about 50 mM oligosaccharide therein; cooling the loaded eukaryotic cells to below their freezing point; and lyophilizing the cooled eukaryotic cells. Lyophilizing preferably is conducted so as to leave a residual water content of less than about 0.40 gram of water per gram of dry weight eukaryotic cells, preferably greater than about 0.15 gram of water per gram of dry weight eukaryotic cells, but more preferably less than about 0.40 gram of water per gram of dry weight of eukaryotic cells.

Further aspects of embodiments of the present invention include a process for increasing the loading efficiency of an oligosaccharide into eukaryotic cells. The process comprises providing eukaryotic cells having a first phase transition temperature range and a second phase transition temperature range (e.g., a temperature greater than about 25°C, such as from about 30°C to less than about 50°C) which is greater than the first phase transition temperature range; disposing the eukaryotic cells in an oligosaccharide solution for loading an oligosaccharide (e.g., trehalose) into the eukaryotic cells; and heating the oligosaccharide solution to the second phase transition temperature range to increase the loading efficiency of the oligosaccharide into the eukaryotic cells. The process additionally comprises taking up external oligosaccharide via fluid phase endocytosis from the oligosaccharide solution.

The present invention also comprises additional embodiments which include a process for increasing the cooperativity of a phase transition of an erythrocytic cell comprising providing an erythrocytic cell having an alcohol (e.g. a sterol) and a phase

transition, and removing at least a portion of the alcohol from the erythrocytic cell to increase the cooperativity of the phase transition of the erythrocytic cell. The erythrocytic cell preferably comprises an erythrocytic membrane including the alcohol and the phase transition. Another embodiment of the present invention provides a process for producing a phase transition temperature range in an erythrocytic cell comprising providing an erythrocytic cell including an alcohol and at least two phase transition temperature ranges, and removing at least a portion of the alcohol from the erythrocytic cell to produce an erythrocytic cell having at least three phase transition temperature ranges. The erythrocytic cell for this feature or aspect of the invention preferably includes an erythrocytic membrane including at least a portion of the alcohol and at least a portion of the two phase transition temperature ranges. After the erythrocytic cell is produced, the produced erythrocytic cell preferably comprises the erythrocytic membrane including at least a portion of the three phase transition temperature ranges after removal of at least a portion of the alcohol.

A further embodiment of the present invention provides a process for loading an oligosaccharide into erythrocytic cells comprising providing erythrocytic cells having an alcohol (e.g. a sterol); removing at least a portion of the alcohol from the erythrocytic cells to produce erythrocytic cells having a phase transition temperature range selected from the group of temperature ranges consisting of a low phase transition temperature range, an intermediate phase transition temperature range, and a high phase transition temperature range; and disposing the erythrocytic cells in an oligosaccharide solution for loading an oligosaccharide (e.g., trehalose) into the

erythrocytic cells. The oligosaccharide solution preferably includes a temperature in a range that approximates the range of temperatures for the phase transition temperature range. The process for loading the oligosaccharide into the erythrocytic cells may additionally comprise heating the oligosaccharide solution, such as to a temperature in the high phase transition temperature range, to increase the loading efficiency of the oligosaccharide into the erythrocytic cells. The process may further additionally comprise taking up external oligosaccharide via lipid phase endocytosis from the oligosaccharide solution. The erythrocytic cells do not necessarily include a fixative.

Another embodiment of the present invention provides a process for increasing the survival of dehydrated erythrocytic cells after storage. The process for increasing survival preferably comprises providing erythrocytic cells from a mammalian species (e.g., a human being) and having an alcohol (e.g. a sterol); removing, preferably at least part of, the alcohol from the erythrocytic cells; and loading the erythrocytic cells with a preservative (e.g., an oligosaccharide). The loaded erythrocytic cells are then dehydrated (e.g., by lyophilizing) while maintaining a residual water content in the erythrocytic cells equal to or less than about 0.30 gram of residual water per gram of dry weight erythrocytic cells to increase erythrocytic cell survival upon rehydrating after storage. The process for increasing survival also preferably comprises storing the dehydrated erythrocytic cells having the residual water content equal to or less than about 0.30 gram of residual water per gram of dry weight erythrocytic cells; and rehydrating the stored dehydrated erythrocytic cells with the stored dehydrated erythrocytic cells surviving dehydration and storage. The loading may be without a

fixative and may comprise taking up external oligosaccharide via lipid phase endocytosis from the oligosaccharide solution. The loading may also, or alternatively, comprise incubating the erythrocytic cells with the oligosaccharide solution. The loaded erythrocytic cells may be cooled to a temperature below their freezing point prior to dehydrating the erythrocytic cells. The residual water content of the erythrocytic cells preferably ranges from about 0.00 gram of residual water per gram of dry weight erythrocytic cells to less than about 0.30 gram of residual water per gram of dry weight erythrocytic cells.

A further embodiment of the present invention provides a process of preparing a dehydrated composition comprising providing erythrocytic cells selected from a mammalian species and including an alcohol (e.g. a sterol); loading internally the erythrocytic cells with more than about 10 mM of an oligosaccharide therein to preserve biological properties; cooling the loaded erythrocytic cells to below their freezing point; and lyophilizing the cooled erythrocytic cells. The loading of the erythrocytic cells for this aspect of the invention may comprise incubating the erythrocytic cells with an oligosaccharide solution having the oligosaccharide therein and a temperature in a range of temperatures selected from the group consisting of a low phase transition temperature range, an intermediate phase transition temperature range, and a high phase transition temperature range. The lyophilizing is conducted so as to leave a residual water content of equal to or less than about 0.3 gram water per gram dry weight of erythrocytic cells. Preferably, greater than about 80% of the erythrocytic cells survive dehydration and storage. The process of preparing a dehydrated composition may additionally comprise

prehydrating the erythrocytic cells, and subsequently hydrating the prehydrated erythrocytic cells.

An additional further embodiment of the present invention comprises a process of preparing loaded erythrocytic cells comprising removing at least a portion of an alcohol (e.g. a sterol) from erythrocytic cells to produce erythrocytic cells having at least three phase transition temperature ranges, and loading (e.g., with an oligosaccharide solution) an oligosaccharide into the erythrocytic cells at a temperature in a range of temperatures approximating one of the three phase transition temperature ranges to produce loaded erythrocytic cells. As previously indicated, the loading may comprise incubating the erythrocytic cells with the oligosaccharide solution at a temperature in a range of temperatures approximating one of the three phase transition temperature ranges.

Additional features of the present invention include a solution for loading erythrocytic cells, an erythrocytic cell composition, and a generally dehydrated composition. The solution for loading erythrocytic cells comprises reduced-alcohol (e.g. reduced-sterol) erythrocytic cells having three phase transition temperature ranges, and an oligosaccharide solution containing the reduced-alcohol erythrocytic cells for loading oligosaccharide from the oligosaccharide solution into the reduced-alcohol erythrocytic cells. External oligosaccharide is taken up via lipid phase endocytosis from the oligosaccharide solution at a temperature in a range of temperatures approximating one of the three phase transition temperature ranges. The erythrocytic cell composition comprises reduced-alcohol erythrocytic cells loaded internally with an oligosaccharide from an oligosaccharide solution. Preferably,

the oligosaccharide is loaded from the oligosaccharide solution at a temperature in a range of temperatures selected from the group consisting of a low phase transition temperature range, an intermediate phase transition temperature range, and a high phase transition temperature range. The generally dehydrated composition comprises freeze-dried reduced-alcohol erythrocytic cells effectively loaded internally with at least about 10 mM of the oligosaccharide (e.g., trehalose) therein to preserve biological properties during freeze-drying and rehydration. The amount of the oligosaccharide loaded inside the freeze-dried reduced-alcohol erythrocytic cells may be from about 10 mM to, about 200 mM. The freeze-dried reduced-alcohol erythrocytic cells may comprise less than about 0.30 gram of residual water per gram of dry weight erythrocytic cells to increase erythrocytic cell survival upon rehydrating.

The sterol may comprise a steroid alcohol, preferably a steroid alcohol having at least one side chain having 8 to 10 carbon atoms. Preferably further, the sterol may comprise from 25 to 27 carbon atoms. More preferably, the sterol comprises cholesterol, such as cholesterol having the formula:

The erythrocytic cells preferably comprise erythrocytic membranes respectively including the low phase transition temperature range, the intermediate phase transition, and the high phase transition temperature range. The low phase transition temperature range is greater than about 2°C, such as a temperature greater than about 2°C to a temperature equal to or less than about 20°C. The intermediate phase transition temperature range is preferably greater than about 20°C, such as a temperature greater than about 20°C to a temperature equal to or less than about 30°C. The high phase transition temperature range is preferably greater than about 30°C, such as a



temperature greater than about 30°C to a temperature equal to or less than about 50°C, more preferably from about 30°C to about 40°C, or from about 32°C to about 38°C.

Further embodiments of the present invention provide a method for loading a preservative into blood platelets comprising providing a preservative solution having a preservative (e.g., an oligosaccharide, such as trehalose), water and protein (e.g., albumin); and loading blood platelets with the preservative solution to produce preservative-loaded blood platelets wherein the preservative solution generally has higher glass transition temperatures than glass transition temperatures for a preservative solution having the preservative, water and no protein. The preservative solution in the preservative-loaded blood platelets comprises a gradient of the glass transition temperature (degrees C) to a water content (grams of water per gram of dry weight of preservative and protein) ranging from about 50 to about 900 at a water content of less than about 0.40 grams of water per gram of dry weight of preservative and protein. The glass transition temperature of the preservative solution in the preservative-loaded blood platelets solution increases at a water content of less than about 0.40 grams of water per gram dry weight of preservative and protein. The preservative solution in the preservative-loaded blood platelets has a greater rate of glass transition temperature per water content (weight of water per dry weight of preservative and protein) increase at a water content of less than about 0.25 grams of water per gram dry weight of preservative and protein than at a water content greater than about 0.25 grams of water per gram dry weight of preservative and protein. More specifically, the preservative solution in the preservative-loaded blood platelets has a

greater rate of glass transition temperature per water content (weight of water per dry weight of preservative and protein) increase at a water content of less than about 0.15 grams of water per gram dry weight of preservative and protein than at a water content of greater than about 0.15 grams of water per gram dry weight of preservative and protein.

Additional embodiments of the present invention provide a process for processing blood platelets comprising providing a preservative solution, such as the preservative solution having a preservative, water, and protein. The process additionally comprises suspending blood platelets in the preservative solution at a concentration greater than about  $10^8$  platelets per ml. of preservative solution to produce preservative-loaded blood platelets; freeze-drying the preservative-loaded blood platelets; and recovering at least 75% of the freeze-dried platelets. The preservative solution may comprise from about 60 mM to about 240 mM of the preservative and from about 2% by weight to about 8% by weight of the protein.

Further additional embodiments of the present invention provide a process for preserving protein structure in blood platelets comprising providing a preservative solution having a preservative, water and protein; loading blood platelets with the preservative solution to produce preservative-loaded blood platelets; and dehydrating the preservative-loaded blood platelets while maintaining a residual water content in the blood platelets equal to or less than about 0.30 gram of residual water per gram of dry weight blood platelets to preserve protein structure of the blood platelets upon rehydrating after storage. The process may additionally include storing the dehydrated preservative-loaded blood platelets; and rehydrating the stored dehydrated preservative-loaded blood platelets with water vapor to preserve protein structure of the

blood platelets and to depress phase transition temperature of membrane lipids. The rehydrating of the stored dehydrated preservative-loaded blood platelets with water vapor comprises increasing the water content of the preservative-loaded blood platelets until the preservative-loaded blood platelets have a water content equal to or less than about 0.30 grams of water per gram of dry weight blood platelets. The process may further additionally comprise directly hydrating with bulk water the rehydrated preservative-loaded blood platelets.

Further additional embodiments of the invention also provide a dehydrated composition for mammalian therapy having freeze-dried platelets including a preservative solution for preserving biological properties during freeze-drying and rehydration. The preservative solution includes water, protein, and a preservative, and the platelets are rehydratable so as to have a normal response to at least one agonist. The normal response to at least one agonists includes a response to thrombin in a physiological concentration commencing at thrombin concentrations ranging from about 0.1 U/ml to about 1.0 U/ml, and wherein between thrombin concentrations ranging from about 0.2 U/ml to about 0.70 U/ml, percent(%) aggregation of the rehydrated platelets ranges from about 20% to about 80%. The normal response to at least one agonists may also include a response to ristocetin in a physiological concentration commencing at ristocetin concentrations ranging from about 1.0 mg/ml to about 10.0 mg/ml. Between ristocetin concentrations ranging from about 2.0 mg/ml to about 10.0 mg/ml, percent(%) aggregation of the rehydrated platelets typically ranges from about 10% to about 100%. Between ristocetin concentrations ranging from about 3.5 mg/ml to about 9.0 mg/ml, percent(%) aggregation of the rehydrated platelets typically ranges from about 40% to about 90%, and between ristocetin concentrations

ranging from about 4.0 mg/ml to about 7.0 mg/ml, percent(%) aggregation of the rehydrated platelets ranges from about 60% to about 80%.

These provisions together with the various ancillary provisions and features which will become apparent to those skilled in the art as the following description proceeds, are attained by the processes, platelets, eukaryotic cells, and erythrocytic cells of the present invention, preferred embodiments thereof being shown with reference to the accompanying drawings, by way of example only, wherein:

### Brief Description of the Drawings

In the drawings:

Figure 1 graphically illustrates the loading efficiency of trehalose plotted versus incubation temperature of human platelets;

Figure 2 graphically illustrates the loading efficiency (cytosolic concentration divided by the extracellular concentration, the sum multiplied by 100) following incubation as a function of incubation time;

Figure 3 graphically illustrates the internal trehalose concentration of human platelets versus external trehalose concentration as a function of temperature at a constant incubation or loading time;

Figure 4 graphically illustrates the loading efficiency of trehalose into human platelets as a function of external trehalose concentration;

Figure 5 graphically illustrates the recovery of platelet embodiments after lyophilization and direct rehydration with various concentrations of trehalose in the drying buffer, and in a combination of 30 mM trehalose and one percent HSA in the drying buffer;

Figure 6 graphically illustrates the uptake of FITC dextran versus the external concentration compared with that of the marker, LYCH (with an incubation time of four hours);

Figure 7 graphically illustrates the effect of prehydration on optical density of platelets;

Figure 8 illustrates the response of 500  $\mu$ l platelets solution (with a platelet concentration of  $0.5 \times 10^8$  cells/ml)

that was transferred to aggregation vials, thrombin added (1U/ml) to each sample, and the samples stirred for three minutes at 37°C, where panel (A) are the prior art platelets and panel (B) are the inventive platelets;

Figure 9 graphically illustrates clot formation where the absorbance falls sharply upon addition of thrombin (1 U/ml) and the platelet concentration drops from  $250 \times 10^6$  platelets/ml to below  $2 \times 10^6$  platelets/ml after three minutes for the inventive platelets;

Figure 10 is a graph illustrating temperatures for membrane phase transition in hydrated mesenchymal stem cells by Fourier transform infrared (FTIR) spectroscopy, with the solid line graph indicating the first derivative of the set of data shown in filled circles;

Figure 11 is a graph representing LYCH loading of mesenchymal stem cells as monitored by fluorescence spectroscopy (filled circles points) and viability as monitored by trypan blue exclusion (filled squares points);

Figures 12A-12B are micrographs of human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope 30 minutes following LYCH-loading, with Fig. 12A showing phase contrast images and all cells intact and Fig. 12B showing fluorescent images for the same cells of Fig. 12A and the LYCH uptake after 30 minutes;

Figures 12C-12D are micrographs of human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope 1 hour following LYCH-loading, with Fig. 12C showing phase contrast images and all cells intact and Fig. 12D showing fluorescent images for the same cells of Fig. 12C and the LYCH uptake after 1 hour;

Figures 12E-12F are micrographs of human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope 2 hours following LYCH-loading, with Fig. 12E showing phase contrast images and all cells intact and Fig. 12F showing fluorescent images for the same cells of Fig. 12E and the LYCH uptake after 2 hours;

Figures 12G-12H are micrographs of human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope 3.5 hours following LYCH-loading, with Fig. 12G showing phase contrast images and all cells intact and Fig. 12H showing fluorescent images for the same cells of Fig. 12G and the LYCH uptake after 3.5 hours;

Figures 12I-12J are micrographs of a control sample (cells incubated in the absence of LYCH) of human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope and having no LYCH-loading of the cells, with Fig. 12I showing phase contrast images and all cells intact and Fig. 12J showing no fluorescent images for the same cells of Fig. 12I because the fluorescence is specific to LYCH and does not correspond to auto-fluorescence from the human mesenchymal stem cells;

Figure 13 is a graph illustrating growth curves for mesenchymal stem cells in the presence or absence of 90 mM trehalose with the open triangle data representing cells grown in standard medium for 24 hours, after which 90 mM trehalose was added;

Figure 14A is a micrograph at a 100X magnification of healthy mesenchymal stem cell culture prior to harvest by trypsinization;

Figure 14B is a micrograph at a 320X magnification of the healthy mesenchymal stem cell culture of Fig. 14A prior to harvest by trypsinization;

Figure 15A is a 100X magnified image of dry lyophilization "cake" of mesenchymal stem cells encased in strands of matrix containing trehalose and BSA;

Figure 15B is a 100X magnified image of prehydrated lyophilization "cake" of mesenchymal stem cells encased in strands of matrix containing trehalose and BSA;

Figure 16A is a micrograph of mesenchymal stem cells magnified 100X following freeze-drying and rehydration;

Figure 16B is a micrograph of mesenchymal stem cells magnified 400X following freezedrying and rehydration;

Figure 16C is a micrograph of mesenchymal stem cells magnified 400X following freezedrying, initial prehydration, and rehydration;

Figure 17A is a micrograph of mesenchymal stem cells from a prehydrated sample at two days post rehydration and illustrating an attached cell and beginning to show characteristic stretched morphology;

Figure 17B is a micrograph of mesenchymal stem cells from a prehydrated sample at five days post rehydration, with nuclei clearly visible in several of the cells;

Figure 18A is a micrograph at 100X magnification of epithelial 293H cells freeze-dried in trehalose, with the cells remaining whole and round, closely resembling their native hydrated state;



Figure 18B is an enlarged view of the dashed square cell field in Fig. 18A with the arrows identifying exceptionally preserved cells;

Figure 19A is a micrograph at 400X magnification of epithelial 293H cells freeze-dried in trehalose; and showing two 293H cells imbedded within a freeze-drying matrix composed of trehalose, albumin, and salts, with the cells appearing whole, round, and completely engulfed within the matrix;

Figure 19B is an enlarged view of the dashed square cell field in Fig. 19A with two cells respectively identified by an arrow;

Figure 20A is a micrograph at 100X magnification of epithelial 293H cells after prehydration (45 min @ 100% RH) and rehydration (1:3 ratio of H<sub>2</sub>O:Growth Medium), and showing a high number of intact, refractile cells;

Figure 20B is an enlarged view of the dashed square cell field in Fig. 20A;

Figure 21A is a micrograph at 320X magnification of epithelial 293H cells 24 hours following rehydration, with refractile whole cells still visible;

Figure 21B is an enlarged view of the dashed square cell field in Fig. 21A with a refractile cell marked by an arrow;

Figure 22A is a graph of cell survival (% control) of trehalose loaded epithelial 293H cells as a function of residual water content measured by trypan blue exclusion;

Figure 22B is another graph of cell survival (% control) of trehalose loaded epithelial 293H cells as a function of residual water content measured by trypan blue exclusion;

Figure 23 is a graph of the residual water content of epithelial 293H cells versus time (minutes) during freeze-drying in a vacuum;

Figure 24 is a graph of wave number versus temperature plot of the CH<sub>2</sub> symmetric stretching mode of erythrocytes from a first blood donor, along with first derivatives of the wave number versus temperature plots;

Figure 25 is a graph of wave number versus temperature plot of the CH<sub>2</sub> symmetric stretching mode of erythrocytes from a second blood donor, along with first derivatives of the wave number versus temperature plots;

Figure 26 is a graph of wave number versus temperature plot of the CH<sub>2</sub> symmetric stretching mode of erythrocytes from a third blood donor, along with first derivatives of the wavenumber versus temperature plots;

Figure 27 is a graph of wavenumber versus temperature plots of control (open circles) and M $\beta$ CD treated erythrocytes (filled circles), along with first derivatives of the wavenumber versus temperature plots (solid lines correspond to M $\beta$ CD treated cells, and dotted lines correspond to control cells);

Figure 28 is a graph of FTIR analysis of the CH<sub>2</sub> stretching region of erythrocytes at 4°C (solid lines) and 37°C (dotted lines), and an absorbance spectra in the 3000-2800 cm<sup>-1</sup> spectral region;

Figure 29 is a graph of FTIR analysis of the CH<sub>2</sub> stretching region of erythrocytes at 4°C (solid lines) and 37°C (dotted lines), with the protein band at 2880 cm<sup>-1</sup> and the lipid band at 2855 cm<sup>-1</sup> being resolved after taking the second derivative of the absorbance spectra;

Figure 30 is a graph of FTIR analysis of the  $\text{CH}_2$  stretching region of erythrocytes at  $4^\circ\text{C}$  (solid lines) and  $37^\circ\text{C}$  (dotted lines), with an inverted second derivative spectra in the  $2890\text{--}2835\text{ cm}^{-1}$  region showing that only the lipid band shifts with temperature;

Figure 31 is a graph illustrating the thermotropic response of the symmetric  $\text{CH}_2$  vibration (filled circles) arising from endogenous lipids, and the symmetric  $\text{CH}_3$  stretch vibration (open circles) arising from endogenous lipids and proteins in intact erythrocytes;

Figure 32 is a graph of wave number versus temperature plot of the  $\text{CH}_2$  symmetric stretching mode of erythrocytes from a first blood donor, along with first derivatives of the wave number versus temperature plots;

Figure 33 is a graph of wave number versus temperature plot of the  $\text{CH}_2$  symmetric stretching mode of erythrocytes from a second blood donor, along with first derivatives of the wave number versus temperature plots;

Figure 34 is a graph of wave number versus temperature plot of the  $\text{CH}_2$  symmetric stretching mode of erythrocytes from a third blood donor, along with first derivatives of the wave number versus temperature plots;

Figure 35 is a graph of wave number versus temperature plots of control (open circles) and M $\beta$ CD treated erythrocytes (filled circles), along with first derivatives of the wave number versus temperature plots (solid lines correspond to M $\beta$ CD treated cells, and dotted lines correspond to control cells);

Figure 36 is a graph of wavenumber versus temperature plots of control ghosts (open circles) and M $\beta$ CD treated ghosts (filled circles), along with first derivatives of the wavenumber versus

temperature plots (solid lines correspond to M $\beta$ CD treated cells, and dotted lines correspond to control cells);

Figure 37 is a graph illustrating the effect (e.g., storage time) of cold storage on erythrocyte membranes versus the wavenumber of the lipid CH<sub>2</sub> stretch vibration at 4°C during storage at 4°C;

Figure 38 is a graph illustrating wavenumber versus temperature plots immediately after isolation (dotted line), after 1 day storage (broken line), and after 5 days storage (solid line) at 4°C;

Figure 39 is an enlarged view of dil-C<sub>18</sub> labeled erythrocytes distribution after 4 days storage at 4°C, illustrating that the dye remained homogeneously distributed in erythrocyte membranes during cold storage;

Figure 40 is a graph representing a state diagram (e.g., glass transition temperature vs. water content) for trehalose alone and for trehalose/albumin (1/1, wt/wt);

Figure 41 is a graph (recovery (%) vs. cell count (#/ml)) illustrating the effects of increasing trehalose and albumin concentrations on survival of freeze-dried blood platelets;

Figure 42 is a graph (relative cell count (%) vs. time (days) illustrating the stability of platelets in the freeze-dried state and suggesting that the shelf life for at least partially active platelets will be at least about two (2) years;

Figure 43 is a graph (relative percentage vs. volume) illustrating the effects of initial prehydration (followed by rehydration) of platelet volume, and the effects of direct rehydration of platelet volume;

Figure 44 is a graph illustrating the effects of prehydration (over water vapor) on phase behavior of freeze-dried platelets;

Figure 45 is a graph illustrating the effects of prehydration (over water vapor) on the cooperativity of phase transition;

Figure 46 is a graph illustrating the effects of prehydration and/or direct rehydration on protein secondary structure in freeze-dried platelets, with direct rehydration significantly altering protein secondary structure relative to controls;

Figure 47 is a graph illustrating the effects of prehydration and/or direct rehydration on protein secondary structure in freeze-dried platelets, with prehydration returning protein secondary structure to control levels before direct rehydration in bulk water;

Figure 48 is a graph (time vs. transmittance (%)) illustrating aggregometry traces for fresh control and freeze-dried (and rehydrated) platelets;

Figure 49 is a graph (thrombin vs. aggregation (%)) illustrating thrombin dose-response curves for control and rehydrated platelets;

Figure 50 is a graph (ristocetin vs. aggregation (%)) illustrating ristocetin dose-response curves for control and rehydrated platelets;

Figure 51 illustrates a collagen dose response curve for fresh platelets; and

Figure 52 illustrates a collagen dose response curve for freeze-dried rehydrated platelets.

Detailed Description of Preferred Embodiments of the Invention

Compositions and embodiments of the invention include platelets that have been manipulated (e.g. by freeze-drying) or modified (e.g. loaded with drugs), and that are useful for therapeutic applications, particularly for platelet transfusion therapy, as surgical or hemostasis aids, such as wound dressings, bandages, and as sutures, and as drug-delivery vehicles. As has been known, human platelets have a phase transition between 12°C and 20°C. We have found that platelets have a second phase transition between 30°C and 37°C. Our discovery of this second phase transition temperature range suggests the possible use of platelets as vehicles for drug delivery because we can load platelets with various useful therapeutic agents without causing abnormalities that interfere with normal platelet responses due to changes, such as in the platelet outer membranes.

For example, platelets may be loaded with anti-thrombic drugs, such as tissue plasminogen activator (TPA) so that the platelets will collect at the site of a thrombus, as in an heart attack, and release the "clot busting" drug or drugs that are encapsulated and have been targeted by the platelets. Antibiotics can also be encapsulated by the platelets, since lipopolysaccharides produced by bacteria attract platelets. Antibiotic loaded platelets will bring the selected antibiotics to the site of inflammation. Other drugs that can be loaded include anti mitotic agents and anti-angiogenic agents. Since platelets circulate in newly formed vessels associated with tumors, they could deliver anti-mitotic drugs in a localized fashion, and likely platelets circulating in the neovasculature of tumors can deposit anti-angiogenic drugs so as to block the blood supply to tumors. Thus, platelets loaded with a selected

drug in accordance with this invention can be prepared and used for therapeutic applications. The drug-loaded platelets are particularly contemplated for blood-borne drug delivery, such as where the selected drug is targeted to a site of platelet-mediated forming thrombi or vascular injury. The so-loaded platelets have a normal response to at least one agonist, particularly to thrombin. Such platelets can be loaded additionally with trehalose, if preservation by freeze-drying is intended.

The key component for compositions and apparatus of embodiments of the invention, when preservation will be by freeze-drying, is a lyoprotectant, preferably an oligosaccharide, more preferably trehalose, because we have found that platelets that are effectively loaded with trehalose preserve biological properties during freeze-drying (and rehydration). This preservation of biological properties, such as the normal clotting response in combination with thrombin, is necessary so that the platelets following preservation can be successfully used in a variety of therapeutic applications.

Normal hemostasis is a sequence of interactions in which blood platelets contribute, beginning with adhesion of platelets to an injured vessel wall. The platelets form an aggregate that accelerates coagulation. A complex, termed the glycoprotein (GP) 1b-IX-V complex, is involved in platelet activation by providing a binding site on the platelet surface for the potent agonist,  $\alpha$ -thrombin.  $\alpha$ -thrombin is a serine protease that is released from damaged tissue. Thus, it is important that the manipulations and modifications in accordance with this invention do not

activate the platelets. Further, it is normally preferred that the platelets be in a resting state. Otherwise, the platelets will activate.

Although for most contemplated therapeutic applications the clotting response to thrombin is key, the inventive freeze-dried platelets after rehydration will also respond to other agonists besides thrombin. These include collagen, ristocetin, and ADP (adenosine diphosphate), all of which are normal platelet agonists. These other agonists typically pertain to specific receptors on the platelet's surface.

Broadly, the preparation of preserved platelets in accordance with the invention comprises the steps of providing a source of platelets, loading the platelets with a protective oligosaccharide at a temperature above about 25°C and less than about 40°C, cooling the loaded platelets to below -32°C, and lyophilizing the platelets.

In order to provide a source of platelets suitable for the inventive preservation process, the platelets are preferably isolated from whole blood. Thus, platelets used in this invention preferably have had other blood components (erythrocytes and leukocytes) removed prior to freeze-drying. The removal of other blood components may be by procedures well known to the art, which typically involve a centrifugation step.

The amount of the preferred trehalose loaded inside the inventive platelets is from about 10 mM to about 50 mM, and is achieved by incubating the platelets to preserve biological properties during freeze-drying with a trehalose solution that has up to about 50 mM trehalose therein. Higher concentrations of trehalose during incubation are not preferred, as will be more fully explained later. The effective loading of trehalose



is also accomplished by means of using an elevated temperature of from greater than about 25° C to less than about 40° C, more preferably from about 30°C to less than about 40°C, most preferably about 37°C. This is due to the discovery of the second phase transition for platelets. As can be seen by Fig. 1, the trehalose loading efficiency begins a steep slope increase at incubation temperatures above about 25°C up to about 40°C. The trehalose concentration in the exterior solution (that is, the loading buffer) and the temperature during incubation together lead to a trehalose uptake that seems to occur primarily through fluid phase endocytosis (that is, pinocytosis). Pinocytosed vesicles lyse over time, which results in a homogeneous distribution of trehalose in the platelets, does not activate the platelets, and can be applied for large scale production. Fig. 2 illustrates the trehalose loading efficiency as a function of incubation time.

As may be gathered from various of the figures, in preparing particularly preferred embodiments, platelets may be loaded with trehalose by incubation at 37°C for about four hours. The trehalose concentration in the loading buffer is preferably 35 mM, which results in an intracellular trehalose concentration of around 20 mM, but in any event is most preferably not greater than about 50 mM trehalose. At trehalose concentrations below about 50 mM, platelets have a normal morphological appearance.

Human platelets have a phase transition between 12°C and 20°C. We found relatively poor loading when the platelets were chilled through the phase transition. Thus, in practicing the method described by U.S. Patent No. 5,827,741, of which some of us are coinventors, only a relatively modest amount of trehalose may be loaded into platelets.

In this application, we have further investigated the phase transition in platelets and have found a second phase transition between 30°C and 37°C. We believe that the excellent loading we obtain at about 37°C is in some way related to this second phase transition. Without being limited by theory, we also believe that pinocytosis is involved, but it may be that the second phase transition itself stimulates the pinocytosis at high temperatures. It may be that other oligosaccharides when loaded in this second phase transition in amounts analogous to trehalose could have similar effects.

In any case, it is fortuitous that the loading can be done at elevated temperatures in view of the fact that chilling platelets slowly -- a requirement for using the first, or lower, phase transition between 20° C and 12° C to introduce trehalose -- is well known to activate them (Tablin et al., *J. Cell. Physiol.*, 168, 305313, 1996). Our relatively high temperature loading, regardless of the mechanism, is thus unexpectedly advantageous both by providing increased loading as well as surprisingly, obviating the activation problem.

Turning to Fig. 6, one sees that we have loaded other, larger molecules into the platelets. In Fig. 6 an illustrative large molecule (FITC dextran) was loaded into the platelets. This illustrates that a wide variety of water-soluble, therapeutic agents can be loaded into the platelets by utilizing the second phase transition, as we have shown may be done with trehalose and with FITC dextran, while still maintaining characteristic platelet surface receptors and avoiding platelet activation.

We have achieved loading efficiencies by practicing the invention with values as high as 61% after four hours incubation. The plateau is not yet reached after four hours. The

high loading efficiency of trehalose is a strong indication that the trehalose is homogeneously distributed rather than located in pinocytosed vesicles, and we expect similar results for loading other therapeutic agents. A loading efficiency of 61% in an external concentration of 25 mM corresponds to a cytosolic concentration of 15 mM. If trehalose were only located in endosomes of 0.1 micrometer, the vesiculation number would be more than 1000. It is unlikely that such a high number of vesicles would be present in platelets next to the other platelet organelles. We therefore believe that the pinocytosed vesicles lyse in the cytoplasm. This results in a homogeneous distribution of trehalose rather than punctuated loading in small vesicles. It is also possible that the trehalose is crossing the membrane due to the phase transition between 30°C and 37°C.

We have found that the endocytotic uptake route is blocked at sugar concentrations above 0.1 M. Consequently, we prefer not to use sugar concentrations higher than about 50 mM in the loading buffer, because at some point above this value we have found swelling and morphological changes of the platelets. Thus, we have found that platelets become swollen after four hours incubation at 37 °C in 75 mM trehalose. Further, at concentrations higher than 50 mM the internal trehalose concentration begins to decrease. By contrast to the present invention, the platelet method taught by Spargo et al., U.S. Patent No. 5,736,313, loads with carbohydrate in the range beginning at about 100 mM and going up to 1.5 M. As noted, we find a high concentration of loading buffer, at least with trehalose, to lead to swelling and morphological changes.

The effective loading of platelets with trehalose is preferably conducted by incubating for at least about two hours,

preferably for at least about four hours. After this loading, then the platelets are cooled to below their freezing point and lyophilized.

Before freezing, the platelets should be placed into a resting state. If not in the resting state, platelets would likely activate. In order to place the platelets in a resting state, a variety of suitable agents, such as calcium channel blockers, may be used. For example, solutions of adenine, adenosine or iloprost are suitable for this purpose. Another suitable agent is PGE<sub>1</sub> (prostaglandin E<sub>1</sub>). It is important that the platelets are not swollen and are completely in the resting state prior to drying. The more they are activated, the more they will be damaged during freeze-drying.

After the platelets have been effectively loaded with trehalose and are in a resting state, then the loading buffer is removed and the platelets are contacted with a drying buffer.

The drying buffer should include trehalose, preferably in amounts up to about 100 mM. The trehalose in the drying buffer assists in spatially separating the platelet as well as stabilizing the platelet membranes on the exterior. The drying buffer preferably also includes a bulking agent (to further separate the platelets). Albumin may serve as a bulking agent, but other polymers may be used with the same effect. If albumin is used, it is preferably from the same species as the platelets. Suitable other polymers, for example, are water-soluble polymers such as HES (hydroxy ethyl starch) and dextran.

The trehalose loaded platelets in drying buffer are then cooled to a temperature below about -32°C. A cooling, that is, freezing, rate is preferably between -30°C and -1°C/min. and more preferably between about -2°C/min to -5°C/min.

Thus, by the practice of embodiments of the present invention, the addition of a protein (e.g., albumin) to a preservative solution (e.g., an oligosaccharide, preferably trehalose, solution) provides for improved and/or optimal survival of preservative (oligosaccharide)-loaded platelets during the freeze-drying and rehydration process. It is suggested that the protein (e.g., the albumin) serves as a bulking agent, physically separating the platelets or cells without contributing significantly to the osmotic pressure of the solution. It is further suggested that the protein (e.g., albumin) requirement is species-specific; that is, if bovine albumin is used with human platelets, for example, the platelets are activated by the bovine (i.e., the foreign) protein. Therefore, the protein (e.g., albumin) is preferably obtained from the same species from which the platelets were obtained.

Osmotic pressure when referred to herein is understood to mean the pressure produced by or associated with osmosis (i.e., the movement of a solvent through a semi-permeable membrane (as of a living cell) into a solution of higher solute concentration that tends to equalize the concentrations of solute on the two sides of the membrane). Osmotic pressure is typically dependent on molar concentration and absolute temperature, such as the maximum pressure that develops in a solution separated from a solvent by a membrane permeable only to the solvent, or the pressure that must be applied to a solution to just prevent osmosis.

Protein when referred to herein means any suitable protein (e.g., simple or conjugated protein), including any complex, high polymer containing carbon, hydrogen, oxygen, nitrogen, and usually sulfur, and composed of chains of amino acids connected by peptide linkages.

Albumin when referred to herein means any suitable albumin (e.g., bovine albumin), including any of a group of water-soluble proteins of wide occurrence in such natural products as milk (lactalbumin), blood serum, eggs (ovalbumin).

Glass transition temperature ( $T_g$ ) when referred to herein means the temperature at which an amorphous matter or material (such as glass, a polymer, blood platelets, or a preservative-protein (e.g., a trehalose-albumin solution)) changes from a brittle vitreous (glassy) state to a plastic state (i.e., a state where the material is capable of being molded or being deformed continuously in any direction without rupture).

In another embodiment of the invention, it has been discovered that protein alters the physical properties of a preservative solution. More specifically, it has been discovered that the protein albumin alters the physical properties of the oligosaccharide trehalose solution as broadly illustrated in Figure. 40, which are graphs representing state diagrams (e.g., glass transition temperature vs. water content) for trehalose:water mixtures and for trehalose-albumin(1/1, wt/wt):water mixtures. A state diagram is a measure of the glass transition temperatures ( $T_g$ ) for the respective mixtures (i.e., the various preservative:water mixtures and the various preservative-protein:water mixtures). The state diagrams broadly illustrated in Figure 40 were obtained by respectively freeze-drying trehalose alone and freeze-drying a 1:1 (wt:wt) trehalose:albumin solution, and then adding back known amounts of water to the respective freeze-dried solutions. The glass transition temperatures for the trehalose:water solution and for the trehalose/albumin:solution were measured by a differential scanning calorimeter with conventional methods.

Long-term stability of matter is improved in the vitreous (glassy) state. Thus, long-term stability of blood platelets, cells, or the like, improves in the glassy state, more specifically when loaded with a preservative having an increased or higher glass transition temperature ( $T_g$ ). Thus, long-term stability for blood platelets, cells, or the like, is improved when maintained in a glassy state.

Generally, an elevated  $T_g$  is distinctly advantageous for long term stability. As illustrated in Figure 40, it has been discovered that albumin elevates significantly the  $T_g$  of trehalose at the water contents indicated in Figure 40. Only at the very lowest water contents was the  $T_g$  not elevated significantly.

In an embodiment of the present invention, blood platelets, cells, or the like (herein for all embodiments "blood platelets") are loaded with a preservative solution to produce preservative-loaded blood platelets. The preservative solution includes a preservative, protein, and water. The preservative may be any suitable preservative, preferably a preservative comprising an oligosaccharide, such as trehalose. The protein may be any suitable protein, preferably a protein comprising albumin (e.g., bovine albumin). The preservative solution comprises the preservative and protein in any suitable mixing or combination ratio, such as from about 0.20 to about 2.00 parts by weight of the preservative to about 1.00 part by weight of the protein (e.g., from about 0.25 grams to about 1.75 grams of preservative per each gram of protein), preferably from about 0.50 to about 1.50 parts by weight of the preservative to about 1.00 part by weight of the protein, more preferably from about 0.75 to about 1.25 parts by weight of the preservative to about 1.00 part by weight of the protein, and most preferably from about 0.90 to about 1.10 parts by weight of the preservative to

about 1.00 part by weight of the protein (e.g., a ratio of about 1 part by weight of preservative to about 1 part by weight of protein). Stated alternatively, but not by way of limitation, the preservative solution comprises from about 60 mM to about 240 mM of the preservative and from about 2% by weight to about 8% by weight of the protein, preferably from about 100 mM to about 200 mM of the preservative and from about 3% by weight to about 7 % by weight of the protein, more preferably from about 125 mM to about 175 mM of the preservative and from about 4 % by weight to about 6% by weight of the protein (e.g., from about 140 mM to about 160 mM of the preservative (or about 150 mM preservative)) and from about 4.5 % by weight to about 5.5 % by weight (or about 5 % by weight) of the protein.

The preservative solution in the preservative-solution loaded blood platelets generally has higher glass transition temperatures than glass transition glass temperatures for a preservative solution having the preservative, water and no protein. The preservative solution has a gradient of the glass transition temperature ( degrees C) to a water content (grams of water per gram of dry weight of preservative and protein) ranging from about 50 to about 900 at a water content of less than about 0.40 grams water per gram of dry weight of preservative and protein. Thus, preservative-loaded blood platelets have a gradient of the glass transition temperature ( degrees C) to a water content (grams of water per gram of dry weight of blood platelets) ranging from about 50 to about 900 at a water content of less than about 0.40 grams water per gram of dry weight of blood platelets. Because the glass transition temperature of the preservative solution increases at a water content of less than about 0.40 (more particularly less than about 0.25) grams of water per gram dry weight of preservative and protein as broadly illustrated in Figure 40, the glass



transition temperature of the preservative-loaded blood platelets increases at a water content of less than about 0.40 (more particularly less than about 0.25) grams of water per gram dry weight of preservative and protein. Stated alternatively, the preservative-loaded blood platelets generally have higher glass transition temperatures at a water content (weight of water per dry weight of blood platelets) of less than about 0.40 grams of water per gram dry weight of blood platelets.

In another embodiment of the invention and as broadly illustrated in Figure 40, the preservative solution in the preservative-loaded blood platelets comprises a greater rate of glass transition temperature per water content (weight of water per dry weight of preservative and protein) increase at a water content of less than about 0.25 grams of water per gram dry weight of preservative and protein than at a water content greater than about 0.25 grams of water per gram dry weight of preservative and protein, more specifically at a water content of less than about 0.15 grams of water per gram dry weight of preservative and protein than at a water content of greater than about 0.15 grams of water per gram dry weight of preservative and protein. Thus, any blood platelets loaded with the preservative solution would have a greater rate of glass transition temperature per water content (weight of water per dry weight of blood platelets) increase at a water content of less than about 0.25 grams of water per gram dry weight of blood platelets than at a water content greater than about 0.25 grams of water per gram dry weight of blood platelets, more specifically at a water content of less than about 0.15 grams of water per gram dry weight of blood platelets than at a water content of greater than about 0.15 grams of water per gram dry weight of blood platelets. Therefore, the preservative-loaded

blood platelets may comprise a water content ranging from about 0.02 grams of water per gram of dry weight of blood platelets to about 0.40 grams of water per gram of dry weight of blood platelets, more specifically from about 0.15 grams of water per gram of dry weight of blood platelets to about 0.40 grams of water per gram of dry weight of blood platelets.

In additional embodiments of the invention and as broadly illustrated in Figure 40, the preservative solution has one of the following gradients of the glass transition temperature (degrees C) to the water content (grams of water per gram of dry weight of preservative and protein): (i) a gradient ranging from about 50 to about 900 at a water content of less than about 0.30 grams of water per gram of dry weight of preservative and protein; (ii) a gradient ranging from about 50 to about 900 at a water content ranging from about 0.02 to less than about 0.40 grams water per gram of dry weight of preservative and protein; (iii) a gradient ranging from about 100 to about 800 at a water content ranging from about 0.15 to about 0.30 grams of water per gram of dry weight of preservative and protein; (iv) a gradient from about 100 to about 800 at a water content ranging from about 0.15 to about 0.30 grams of water per gram of dry weight of preservative and protein; (v) a gradient from about 50 to about 150 at a water content ranging from about 0.20 to about 0.30 grams of water per gram of dry weight of preservative and protein; (vi) a gradient from about 75 to about 125 at a water content ranging from about 0.20 to about 0.30 grams of water per gram of dry weight of preservative and protein; (vii) a gradient from about 700 to about 900 at a water content ranging from about 0.15 to about 0.20 grams of water per gram of dry weight of preservative and protein; and (viii) a gradient from about 750 to about 850 at a water content ranging from about 0.15 to about

0.20 grams of water per gram of dry weight of preservative and protein.

Correspondingly, therefore, the preservative loaded blood platelets include one of the following gradients of the glass transition temperature (degrees C) to the water content (grams of water per gram of dry weight of blood platelets) (i) a gradient ranging from about 50 to about 900 at a water content of less than about 0.30 grams of water per gram of dry weight of blood platelets; (ii) a gradient ranging from about 50 to about 900 at a water content ranging from about 0.02 to less than about 0.40 grams water per gram of dry weight of blood platelets; (iii) a gradient ranging from about 100 to about 800 at a water content ranging from about 0.15 to about 0.30 grams of water per gram of dry weight of blood platelets; (iv) a gradient from about 100 to about 800 at a water content ranging from about 0.15 to about 0.30 grams of water per gram of dry weight of blood platelets; (v) a gradient from about 50 to about 150 at a water content ranging from about 0.20 to about 0.30 grams of water per gram of dry weight of blood platelets; (vi) a gradient from about 75 to about 125 at a water content ranging from about 0.20 to about 0.30 grams of water per gram of dry weight of blood platelets; (vii) a gradient from about 700 to about 900 at a water content ranging from about 0.15 to about 0.20 grams of water per gram of dry weight of blood platelets; and (viii) a gradient from about 750 to about 850 at a water content ranging from about 0.15 to about 0.20 grams of water per gram of dry weight of blood platelets.

Thus, during processing (e.g., freeze-drying processing) a preservative solution with albumin enters the glassy state at higher water contents. Therefore, correspondingly blood platelets loaded with a preservative solution having albumin will enter the glassy state at higher water contents. These

effects increase the stability in storing blood platelets. By way of example only and referencing Figure 40, assuming that primary drying is done at a sample temperature of  $-50^{\circ}\text{C}$ , samples with albumin vitrify (in the glassy state) at a water content as high as about 0.4 gram  $\text{H}_2\text{O}$ /gram dry wt. As illustrated in Figure 40, samples with albumin will enter into the glassy state at a water content as high as about 0.4 gram  $\text{H}_2\text{O}$ /gram dry wt, while those with trehalose alone (i.e., not having albumin) will not enter the glassy state until the water content falls below about 0.15 gram  $\text{H}_2\text{O}$ /gram dry wt. A water content ranging from about 0.15 gram  $\text{H}_2\text{O}$ /gram dry wt to about 0.4 gram  $\text{H}_2\text{O}$ /gram dry wt is a preferred water content range when the water of hydration of membranes and proteins is being removed (i.e., when membranes and proteins are being dehydrated). In addition, the rate of water removal decelerates sharply in this range of water contents, so the samples with albumin would fall below  $T_g$  earlier or sooner than those without albumin, leading to decreased opportunities for damage.

The lyophilization step is preferably conducted at a temperature below about  $-32^{\circ}\text{C}$ , for example conducted at about  $-40^{\circ}\text{C}$ , and drying may be continued until about 95 weight percent of water has been removed from the platelets. During the initial stages of lyophilization, the pressure is preferably at about  $10 \times 10^{-6}$  torr. As the samples dry, the temperature can be raised to be warmer than  $-32^{\circ}\text{C}$ . Based upon the bulk of the sample, the temperature and the pressure it can be empirically determined what the most efficient temperature values should be in order to maximize the evaporative water loss. Freeze-dried compositions of the invention preferably have less than about 5 weight percent water.

The freeze-dried platelets may be used by themselves, dissolved in a physiologically acceptable solution, or may be a component of a biologically compatible (biocompatible) structure or matrix, which provides a surface on or by which the freeze-dried platelets are carried. The freeze-dried platelets can be, for example, applied as a coating to or impregnated in a wide variety of known and useful materials suitable as biocompatible structures for therapeutic applications. The earlier mentioned U.S. Patent No. 5,902,608, for example, discusses a number of materials useful for surgical aid, wound dressings, bandages, sutures, prosthetic devices, and the like. Sutures, for example, can be monofilament or braided, can be biodegradable or nonbiodegradable, and can be made of materials such as nylon, silk, polyester, cotton, catgut, homopolymers, and copolymers of glycolide and lactide, etc. Polymeric materials can also be cast as a thin film, sterilized, and packaged for use as a wound dressing. Bandages may be made of any suitable substrate material, such as woven or nonwoven cotton or other fabric suitable for application to or over a wound, may optionally include a backing material, and may optionally include one or more adhesive regions on the face surface thereof for securing the bandage over the wound.

The freeze-dried platelets, whether by themselves, as a component of a vial-compatible structure or matrix, and optionally including other dry or freeze-dried components, maybe packaged so as to prevent rehydration until desired. The packaging may be any of the various suitable packagings for therapeutic purposes, such as made from foil, metallized plastic materials, and moisture barrier plastics (e.g. high-density polyethylene or plastic films that have been created with materials such as SiO<sub>x</sub>), cooling the trehalose loaded platelets

to below their freezing point, and lyophilizing the cooled platelets. The trehalose loading includes incubating the platelets at a temperature from greater than about 25°C to less than about 40°C with a trehalose solution having up to about 50 mM trehalose therein. The process of using such a dehydrated composition comprises rehydrating the platelets. The rehydration preferably includes a prehydration step, sufficient to bring the water content of the freeze-dried platelets to between about 20 weight percent and about 50 percent, preferably from about 20 weight percent to about 40 weight percent.

When reconstitution is desired, prehydration of the freeze-dried platelets in moisture saturated air followed by rehydration is preferred. Use of prehydration yields cells with a much more dense appearance and with no balloon cells being present. Prehydrated, previously lyophilized platelets of the invention resemble fresh platelets. This is illustrated, for example, by Fig. 7. As can be seen, the previously freeze-dried platelets can be restored to a condition that looks like fresh platelets.

Before the prehydration step, it is desirable to have diluted the platelets in the drying buffer to prevent aggregation during the prehydration and rehydration. At concentrations below about  $3 \times 10^8$  cells/ml, the ultimate recovery is about 70% with no visible aggregates. Prehydration is preferably conducted in moisture saturated air, most preferably is conducted at about 37°C for about one hour to about three hours. The preferred prehydration step brings the water content of the freeze-dried platelets to between about 20 weight percent to about 50 weight percent.

The prehydrated platelets may then be fully rehydrated. Rehydration may be with any aqueous based solutions, depending

upon the intended application. In one preferred rehydration, we used plasma, which resulted in about 90% recovery.

Since it is frequently desirable to dilute the platelets to prevent aggregation when the freeze-dried platelets are once again hydrated, it may then be desired or necessary for particular clinical applications to concentrate the platelets. Concentration can be by any conventional means, such as by centrifugation. In general, a rehydrated platelet composition will preferably have  $10^6$  to  $10^{11}$  platelets per ml, more preferably  $10^8$  to  $10^{10}$  platelets per ml.

By contrast with the previous attempts at freeze drying platelets, we show here that with a very simple loading, freeze-drying and rehydration protocol one obtains platelets that are morphologically intact after rehydration, and have an identical response to thrombin as do fresh platelets. Moreover, the concentration of thrombin to give this response is a physiological concentration -- 1 U/ml.

For example, Fig. 8, panel (A), illustrates the clot formation for fresh platelets and in panel (B) for platelets that have been preserved and then rehydrated in accordance with this invention. The cell counts that were determined after three minutes exposure to thrombin were zero for both the fresh platelets and the previously freeze-dried platelets of the invention.

Fig. 9 graphically illustrates clotting as measured with an aggregometer. With this instrument one can measure the change in transmittance when a clot is formed. The initial platelet concentration was  $250 \times 10^6$  platelets/ml, and then thrombin (1 U/ml) was added and the clot formation was monitored with the aggregometer. The absorbance fell sharply and the cell count

dropped, to below  $2 \times 10^6$  platelets/ml after three minutes, which was comparable to the results when the test was run with fresh platelets as a control.

Thus, for various embodiments of the invention, platelet or cell counts or concentrations range from about  $10^6$  to about  $10^{11}$  platelets per ml preservative solution. For additional various embodiments of the invention, platelets may be successfully freeze-dried at concentrations greater than about  $10^8$  platelets per ml preservative, such as from about  $10^8$  platelets per ml preservative to about  $10^{10}$  platelets per ml, more specifically such as from about  $0.5 \times 10^9$  platelets per ml preservative solution to about  $10.0 \times 10^9$  platelets per ml preservative solution including at least about  $5 \times 10^9$  platelets per ml preservative solution.

It has been discovered that at least about 75% of the freeze-dried platelets survive freeze-drying, or may be recovered through hydration. More specifically, at least about 85%, including at least about 90%, of the freeze-dried platelets survive the preservation procedures described herein, and/or may be recovered through hydration. In an embodiment of the invention, the per cent (%) surviving may be a per cent (%) of the number of platelets. It has been further discovered that for any particular concentration of platelets, increasing the quantity of protein and/or the quantity of preservative in the preservative solution, increases the survival or recovery of the freeze-dried platelets at that particular concentration of platelets. By way of example only and referencing now Figure 41 wherein there is illustrated a graph (recovery (%) vs. cell count (#/ml)) illustrating the effects of increasing trehalose and albumin concentrations on survival of freeze-dried blood platelets, a preservative solution having 1% by weight trehalose



and 30 mM albumin limited the platelet or cell survival to around  $10^8$  cells/ml of preservative solution. However, when the albumin concentration was increased to 5% by weight and the trehalose concentration was increased to 150 mM, cells or platelets illustrated good survival all the way up to about  $5 \times 10^9$  cells/ml of preservative solution. Preferably the amount of albumin (i.e., the protein) and the amount of preservative (i.e., trehalose) is respectively larger than 1% by weight and 30 mM. Thus, another embodiment of the invention provides for increasing the survival of platelets by increasing the amount of preservative (e.g., trehalose) and/or the amount of protein (e.g., albumin). Preferably, both the amount of preservative and the amount of protein are increased. As indicated and in an embodiment of the invention, but not by way of limitation, the preservative solution comprises, either singularly or in combination, at least about 60 mM preservative and at least about 2 % by weight protein e.g., from about 60 mM to about 240 mM of the preservative and from about 2% by weight to about 8% by weight of the protein), preferably at least about 100 mM preservative and at least about 3 % by weight protein (e.g., from about 100 mM to about 200 mM of the preservative and from about 3% by weight to about 7 % by weight of the protein, more preferably at least about 125 mM preservative preservative and at least about 4 % by weight protein (e.g., from about 125 mM to about 175 mM of the preservative and from about 4 % by weight to about 6% by weight of the protein [e.g., from about 140 mM to about 160 mM of the preservative (or about 150 mM preservative)) and from about 4.5 % by weight to about 5.5 % by weight (or about 5 % by weight) of the protein].

Therefore, an embodiment of the present invention broadly provides a process for processing blood platelets comprising suspending blood platelets in a preservative solution (e.g., a

preservative solution having water, protein and a preservative) at a concentration greater than about  $10^8$  platelets per ml. of preservative solution to produce preservative-loaded blood platelets; freeze-drying the preservative-loaded blood platelets; and recovering at least 75% (including at least 85%) of the freeze-dried platelets. As indicated, the preservative solution may comprise from about 60 mM to about 240 mM of a preservative and from about 2% by weight to about 8% by weight protein. The concentration may range from about  $0.5 \times 10^9$  platelets per ml preservative solution to about  $10.0 \times 10^9$  platelets per ml preservative solution. The process may additionally comprise storing, prior to recovering, the freeze-dried platelets for more than 600 days ( e.g., for about 2 years or longer).

As indicated for various embodiments of the invention, blood platelets which are freeze-dried with trehalose and albumin respond normally to thrombin at physiological levels. These initial measurements and observations were done by visual observation of clotting and by cell shape change. These observations may be amplified and extended to other agonists, including ristocetin and collagen, through the assistance of aggregometry, which is a technique that measures the time course of platelet clot formation by recording optical density. As the platelets clot and fall out of solution, the optical density decreases. This is a standard clinical assay for platelet clotting.

Referring now to Figure 48 there is seen a graph (time vs. transmittance (%)) illustrating aggregometry traces for fresh control and freeze-dried (and rehydrated) platelets. Even though platelet clotting in freeze-dried platelets may not be as effective when compared with the fresh controls (see graph **480** for fresh control platelets vs. graph **484** for rehydrated

platelets), it is nevertheless clear from graph **484** in Figure 48 that the freeze-dried platelets respond to thrombin (e.g., respond at thrombin concentrations at around 1U/ml), and that the response is within the range of normal controls.

In another embodiment of the invention, it has been discovered that freeze-dried platelets respond to agonists (e.g., thrombin) below about 1 U/ml. Referring now to Figure 49, there is seen a graph (thrombin vs. aggregation (%)) illustrating a thrombin dose-response curve **490** for fresh control platelets and a thrombin dose-response curve **494** for rehydrated platelets. From the curve **494** in Figure 49 it may be seen that the clotting response of rehydrated platelets to thrombin clearly commences at thrombin concentrations below 1 U/ml, with maximal aggregation being achieved at approximately 1 U/ml. As illustrated in Figure 49, response to thrombin commences at thrombin concentrations ranging from about 0.1 U/ml to about 1.0 U/ml. Between thrombin concentrations ranging from about 0.2 U/ml to about 0.70 U/ml, percent(%) aggregation of the rehydrated platelets ranges from about 20% to about 80%. Between thrombin concentrations ranging from about 0.35 U/ml to about 0.70 U/ml, percent(%) aggregation of the rehydrated platelets ranges from about 40% to about 80%.

The findings and theories with respect to thrombin may be extended to other agonists, including ristocetin and collagen. Ristocetin is a non-physiological agonist which requires an active conformation of the GpI-V-IX complex in order to obtain binding and subsequent aggregation. This same complex is required for the binding of von Willebrand factor (vWf) to GPIb complex in vivo, and is used as an in vitro assessment of GPIb availability. Referring now to Figure 50, there is seen a graph (ristocetin vs. aggregation (%)) illustrating a ristocetin dose-response curve **502** for fresh control platelets and a ristocetin

dose-response curve **504** for rehydrated platelets. As illustrated in Figure 50, response to ristocetin commences at ristocetin concentrations ranging from about 1.0 mg/ml to about 10.0 mg/ml. Between ristocetin concentrations ranging from about 2.0 mg/ml to about 10.0 mg/ml, percent(%) aggregation of the rehydrated platelets ranges from about 10% to about 100%. Between ristocetin concentrations ranging from about 3.5 mg/ml to about 9.0 mg/ml, percent(%) aggregation of the rehydrated platelets ranges from about 40% to about 90%. Between ristocetin concentrations ranging from about 4.0 mg/ml to about 7.0 mg/ml, percent(%) aggregation of the rehydrated platelets ranges from about 60% to about 80%. The foregoing results demonstrate that rehydrated freeze-dried platelets are able to respond to ristocetin in a dose dependent manner, with aggregation for both fresh control and rehydrated freeze-dried platelets reaching about 100%.

Referring now to Figures 51 and 52 there are illustrated respectively collagen dose response curves, generally illustrated as **510**, for fresh control platelets, and collagen dose response curves, generally illustrated as **520**, for freeze-dried rehydrated platelets. Collagen dose response curves **510** include fresh control platelet curves **512**, **514**, and **516**, respectively representing percent aggregation in response to collagen doses to fresh platelets from three individuals. Collagen dose response curves **520** include rehydrated platelet curves **522**, **524**, and **528**, respectively representing percent aggregation in response to collagen doses to rehydrated platelets from three individuals.

As illustrated in Figure 52, rehydrated platelets respond to collagen. In an embodiment of the invention, extracellular calcium (e.g., 0.5 to 4.0 mM, such as 2.0 mM) may be added in the physiological range. As shown in Figure 51, fresh platelets

have a rapid and complete response to about 10 ug/ml collagen, while freeze-dried rehydrated platelets require from about 50 ug/ml to about 100 ug/ml of collagen to initiate an aggregation response. At approximately 200 ug/ml of collagen the response is maximal at about 40% aggregation.

Although platelets for use in embodiments of this invention preferably have had other blood components removed before freeze-drying, compositions and apparatuses of embodiments of the invention may also include a variety of additional therapeutic agents. For example, particularly for embodiments contemplated in hemostasis applications, antifungal and antibacterial agents are usefully included with the platelets, such as being admixed with the platelets. Embodiments can also include admixtures or compositions including freeze-dried collagen, which provides a thrombogenic surface for the platelets. Other components that can provide a freeze-dried extra-cellular matrix can be used, for example, components composed of proteoglycan. Yet other therapeutic agents that may be included in inventive embodiments are growth factors. When the embodiments include such other components, or admixtures, they are preferably in dry form, and most preferably are also freeze-dried. We also contemplate therapeutic uses of the composition where additional therapeutic agents may be incorporated into or admixed with the platelets in hydrated form. The platelets, as earlier mentioned, can also be prepared as to encapsulate drugs in drug delivery applications. If trehalose is also loaded into the platelet interiors, then such drug encapsulated platelets may be freeze-dried as has been earlier described.

The platelets should be selected of the mammalian species for which treatment is intended (e.g. human, equine, canine,

feline, or endangered species), most preferably human. The injuries to be treated by applying hemostasis aids with the platelets include abrasions, incisions, burns, and may be wounds occurring during surgery of organs or of skin tissue. The platelets of the invention may be applied or delivered to the location of such injury or wound by any suitable means. For example, application of inventive embodiments to burns can encourage the development of scabs, the formation of chemotactic gradients, of matrices for inducing blood vessel growth, and eventually for skin cells to move across and fill in the burn. For transfusion therapy, inventive compositions may be reconstituted (rehydrated) as pharmaceutical formulations and administered to human patients by intravenous injection. Such pharmaceutical formulations may include any aqueous carrier suitable for rehydrating the platelets (e.g., sterile, physiological saline solution, including buffers and other therapeutically active agents that may be included in the reconstituted formulation). For drug delivery, the inventive compositions will typically be administered into the blood stream, such as by i.v.

In additional embodiments of the present invention, it has been discovered that the general findings with respect to platelets are broadly applicable to cells, particularly eukaryotic cells. The term "eukaryotic cell" is used to mean any nucleated cell, i.e., a cell that possesses a nucleus surrounded by a nuclear membrane, as well as any cell that is derived by terminal differentiation from a nucleated cell, even though the derived cell is not nucleated. Examples of the latter are terminally differentiated human red blood cells. Mammalian, and particularly human, eukaryotes are preferred. Suitable mammalian

species include by way of example only, not only human, but also equine, canine, feline, or endangered species.

Thus, compositions and embodiments of the present invention include eukaryotic cells (e.g., mesenchymal stem cells, epithelial 293H cells, etc) that have been manipulated (e.g. by freeze-drying) or modified (e.g. loaded with preservatives) and that are useful for well known therapeutic applications. We have discovered that eukaryotic cells have a first phase transition between about  $-10^{\circ}\text{C}$  and about  $24^{\circ}\text{C}$  and a second phase transition at temperatures commencing with about  $25^{\circ}\text{C}$  and terminating at temperatures of about  $50^{\circ}\text{C}$ . More specifically, we have discovered that eukaryotic cells have a second phase transition at a temperature greater than about  $25^{\circ}\text{C}$ , such as a temperature ranging from a temperature greater than about  $25^{\circ}\text{C}$  to a temperature less than about  $50^{\circ}\text{C}$ , including a temperature ranging from about  $30^{\circ}\text{C}$  to less than about  $50^{\circ}\text{C}$ , more particularly a temperature ranging from about  $30^{\circ}\text{C}$  to about  $40^{\circ}\text{C}$ , most preferably a temperature ranging from about  $32^{\circ}\text{C}$  to about  $38^{\circ}\text{C}$ , such as from about  $34^{\circ}\text{C}$  to about  $37^{\circ}\text{C}$ . Our discovery of this second phase transition suggests improving the preservation of eukaryotic cells by optimizing loading eukaryotic cells with a preservative (.e.g., an oligosaccharide, such as trehalose), and by optimizing the storage and rehydration of eukaryotic cells. We have more specifically discovered that eukaryotic cells, which were loaded with trehalose at the second phase transition temperature range and freeze dried, are viable immediately following rehydration and appear healthy because the membranes are intact and the nuclei are clearly visible and are of normal morphology.

One of the salient components for compositions and apparatus of additional embodiments of the present invention,

when cell preservation will be assisted by freeze-drying, is an oligosaccharide, preferably trehalose, because we have discovered that eukaryotic cells which are effectively loaded with trehalose preserve biological properties during freeze drying (and rehydration). This preservation of biological properties, such as the immediate restoration of viability following rehydration, is necessary so that the eukaryotic cells following preservation can be successfully used in a variety of well known therapeutic applications. Preferably, the preparation of preserved eukaryotic cells in accordance with embodiments of the present invention broadly comprises the steps of providing a source of eukaryotic cells, loading the eukaryotic cells with a protective preservative (e.g., an oligosaccharide) at a temperature above 25°C and less than about 50°C, cooling the loaded eukaryotic cells to below -32°C, and lyophilizing the eukaryotic cells.

The source of the eukaryotic cells may be any suitable source such that the eukaryotic cells may be cultivated in accordance with well known procedures, such as incubating the eukaryotic cells with a suitable serum (e.g., fetal bovine serum). After the eukaryotic cells are cultured, they are subsequently harvested by any conventional procedure, such as by trypsinization, in order to be loaded with a protective preservative. The eukaryotic cells are preferably loaded by growing the eukaryotic cells in a liquid tissue culture medium. The preservative (e.g., an oligosaccharide, such as trehalose) is preferably dissolved in the liquid tissue culture medium, which includes any liquid solution capable of preserving living cells and tissue. Many types of mammalian tissue culture media are known in the literature and available from commercial suppliers, such as Sigma Chemical Company, St. Louis, Mo., USA:



Aldrich Chemical Company, Inc., Milwaukee, Wis., USA; and Gibco BRL Life Technologies, Inc., Grand Island, N.Y., USA. Examples of media that are commercially available are Basal Medium Eagle, CRCM-30 Medium, CMRL Medium-1066, Dulbecco's Modified Eagle's Medium, Fischer's Medium, Glasgow Minimum Essential Medium, Ham's F-10 Medium, Ham's F-12 Medium, High Cell Density Medium, Iscove's Modified Dulbecco's Medium, Leibovitz's L-15 Medium, McCoy's 5A Medium (modified), Medium 199, Minimum Essential Medium Eagle, Alpha Minimum Essential Medium, Earle's Minimum Essential Medium, Medium NCTC 109, Medium NCTC 135, RPMMI-1640 Medium, William's Medium E, Waymouth's MB 752/1 Medium, and Waymouth's MB 705/1 Medium.

When the preservative to be loaded in the eukaryotic cells is trehalose, the actual amount of trehalose dissolved in the liquid tissue culture medium may vary, although considerations of the economical use of materials and labor, and considerations of the cryopreservation protocol, i.e., the choice of procedural steps used for cooling and thawing the eukaryotic cells together with the cooling and thawing rates, may affect the selection of concentration ranges that will provide the most efficient and effective preservation. In the case of trehalose for one embodiment of the present invention, the concentration of trehalose in the cryopreservation medium (i.e., the tissue culture medium plus added trehalose) ranges from about 10mM and about 1.5 M, preferably between about 100mM and about 500mM, in the cryopreservation medium. In another embodiment of the present invention, the concentration of trehalose in the cryopreservation medium ranges from about 10mM to less than about 100mM, such as from about 10mM to about 50mM, in the cryopreservation medium. The concentration of the eukaryotic cells in the cryopreservation medium that will provide optimal results may vary, and the

concentration selected for use in any given procedure will be governed primarily by consideration of economy and efficiency. Effective results will generally be achieved with suspensions containing from about  $10^5$  to about  $10^{10}$  eukaryotic cells per milliliter of cryopreservation medium, preferably from about  $10^6$  to about  $10^9$  eukaryotic cells/mL, and most preferably from about  $10^7$  to about  $10^8$  eukaryotic cells/mL.

The amount of the preferred trehalose loaded inside the eukaryotic cells may be any suitable amount, preferably from about 10 mM to less than about 100 mM, more preferably from about 10 mM to about 90 mM, most preferably from about 10 mM to about 50 mM, and is preferably achieved by incubating the eukaryotic cells to preserve biological properties during freeze-drying with a trehalose solution that has less than about 100 mM trehalose therein. As was found for platelets, higher concentrations of trehalose during incubation are not preferred. The effective loading of trehalose is also accomplished by means of using an elevated temperature of from greater than about  $25^{\circ}\text{C}$  to less than about  $50^{\circ}\text{C}$ , more preferably from about  $30^{\circ}\text{C}$  to less than about  $40^{\circ}\text{C}$ , most preferably about  $35^{\circ}\text{C}$ . This is due to the discovery of the second phase transition for eukaryotic cells. It is believed that the trehalose loading efficiency for eukaryotic cells increase at incubation temperatures above about  $25^{\circ}\text{C}$  up to about  $50^{\circ}\text{C}$ . Thus, it is believed that the Fig. 1 graph for platelets would be applicable for eukaryotic cells when the steep upwardly sloping line in Fig. 1 is extended to an incubation temperature of about  $50^{\circ}\text{C}$ .

The trehalose concentration in the exterior solution (that is, the loading buffer or cryopreservation medium) and the temperature during incubation together lead to a trehalose uptake that occurs primarily through fluid phase endocytosis

(i.e., pinocytosis). Pinocytosed vesicles lyse over time which results in a homogeneous distribution of trehalose in the eukaryotic cells. Without being limited by theory, while we believe that pinocytosis is involved, it may be that the second phase transition itself stimulates the pinocytosis at high temperatures. It is believed that other oligosaccharides when loaded in this second phase transition in amounts analogous to trehalose could have similar effects. It is also believed that the trehalose loading efficiency as a function of incubation time for eukaryotic cells would be comparable to that of platelets. Thus, Fig. 2 would be representative of the trehalose loading efficiency as a function of incubation time for eukaryotic cells.

Lipid phase transitions in the eukaryotic cells are preferably measured by changes in membrane  $\text{CH}_2$  vibrational frequency, using a Fourier transform infrared microscope coupled to an optical bench and equipped with a temperature controller. Samples may be prepared by placing the eukaryotic cells between  $\text{CaF}_2$  windows, and placing the windows and eukaryotic cells in the temperature controller on the microscope stage. All curve fitting may be done by multiple iterations of a least squares algorithm on a microcomputer.

In preparing particularly preferred embodiments of the invention, eukaryotic cells may be loaded with trehalose by incubation at about  $37^\circ\text{C}$  from about four to about twenty-four hours. The trehalose concentration in the loading buffer or cryopreservation medium is preferably about 35 mM, which results in an intracellular trehalose concentration of around 20 mM, but in any event is most preferably not greater than about 50 mM trehalose. At trehalose concentrations below about 50 mM, eukaryotic cells have a normal morphological appearance.

After the eukaryotic cells have been effectively loaded with a preservative (e.g., an oligosaccharide, such as trehalose), then the loading buffer or cryopreservation medium is removed and the eukaryotic cells are contacted with a drying buffer (i.e., a freeze-drying buffer). Drying of eukaryotic cells after preservative loading may be carried out by suspending the eukaryotic cells in a suitable drying solution containing a suitable bulking agent (or drying buffer), such as in any suitable drying solution containing a salt, a starch, or an albumin. The drying buffer preferably also includes the preservative (e.g., trehalose), preferably in amounts up to about 200 mM, more preferably up to about 100 mM. Trehalose in the drying buffer assists in spatially separating the eukaryotic cells as well as stabilizing the eukaryotic membranes on the exterior. The drying buffer preferably also includes a bulking agent (to further separate the eukaryotic cells). As previously indicated, albumin may serve as a bulking agent, but other polymers may be used with the same effect. Suitable other polymers, for example, are water-soluble polymers such as HES and dextran.

The preservative (trehalose) loaded eukaryotic cells in the drying buffer are then cooled to a temperature below about  $-32^{\circ}\text{C}$ . A cooling (i.e. freezing) rate is preferably between  $-30^{\circ}\text{C}$  and  $-1^{\circ}\text{C}/\text{min.}$ , and more preferably between about  $-2^{\circ}\text{C}/\text{min}$  to  $-5^{\circ}\text{C}/\text{min}$ . The lyophilization step is preferably conducted at a temperature below about  $-32^{\circ}\text{C}$ , for example conducted at about  $-40^{\circ}\text{C}$ .

In one embodiment of the present invention, drying may be continued until about 95 weight percent of water has been removed from the eukaryotic cells. During the initial stages of lyophilization, the pressure is preferably at about  $10 \times 10^{-6}$

Torr. As the cell samples dry, the temperature may be raised to be warmer than  $-32^{\circ}\text{C}$ . Based upon the bulk of the cell samples, the temperature, and the pressure, it may be empirically determined what the most efficient temperature values should be in order to maximize the evaporative water loss. For this embodiment of the invention, freeze-dried eukaryotic cell compositions may have less than about 5 weight percent water.

In another embodiment of the invention, drying of the eukaryotic cells is continued until the water content of the eukaryotic cells does not fall below about 0.15 grams of water per gram of dry weight eukaryotic cells, more preferably not below about 0.20 grams of water per gram of dry weight eukaryotic cells. Preferably, the water content of the dried (e.g., freeze-dried) eukaryotic cells is maintained from about 0.20 gram of residual water per gram of dry weight eukaryotic cells to about 0.75 gram of residual water per gram of dry weight eukaryotic cells. For this embodiment of the invention, dehydration does not mean removal of 100% contained water. It has been discovered that by retention of greater than 0.15 gm water per gm dry weight eukaryotic cells, the survival percentage of the eukaryotic cells after removal from the lyophilizer and rehydration is more than about 80%.

Referring now to Fig. 22A there is seen a graph of cell survival (% control) for trehalose loaded epithelial 293H cells as a function of residual water content measured by trypan blue exclusion. Fig. 22A clearly shows that for residual water contents greater than about 0.15 gram of residual water per gram of dry weight eukaryotic cells, cell survival is high (e.g., greater than about 80%), but descends toward zero (0) if less than about 0.15 grams of water per gram of dry weight eukaryotic cells is retained. Fig. 22B is another graph of cell survival (%)

control) of trehalose loaded epithelial 293H cells as a function of residual water content measured by trypan blue exclusion. Fig. 23 is a graph of the water content of epithelial 293H cells vs. time (minutes) of vacuum drying. The results illustrated in Fig. 23 were obtained by loading the epithelial 293H cells with trehalose, then cooling and freezing, and subsequently transferring the cells to a side arm lyophilizer, which permitted selective removal of cell samples one at a time during the freeze-drying process. The cell samples were removed at the indicated time intervals, weighed, and then oven dried to constant weight. The water content at each time point shown in Fig. 23 was calculated from the wet (or water) weight-dry weight difference. The freeze-dried eukaryotic cell compositions for this embodiment of the invention have more than about 0.15 gram of residual water per gram of dry weight eukaryotic cells.

Referring in detail now to Figure 42, there is seen a graph (relative cell count (%) vs. time (days) illustrating the stability of platelets in the freeze-dried state and suggesting that the shelf life for at least partially active platelets will be at least about two (2) years. Platelets were freeze-dried with trehalose and albumin. The samples were freeze dried in vials, which were then flushed with nitrogen and stored in the dark in an effort at inhibiting photo-oxidation. Upon rehydration and as illustrated in Figure 42, there was virtually no loss of platelets at storage times approaching 700 days. The response of these platelets to any of the agonists for embodiments of the present invention is normal.

As previously indicated, freeze-dried platelets may be prehydrated over water vapor in order to increase platelet survival. Referring now to Figure 43 there is seen a graph (relative percentage vs. volume) illustrating the effects of

initial prehydration (followed by rehydration) of platelet volume, and the effects of direct rehydration of platelet volume. More specifically shown in Figure 43 are fresh control graph **430**, prehydrated graph **434**, directly rehydrated graph **436**. Figure 43 more particularly shows that prehydrated cells (see prehydrated graph **434**) have a cell volume very close to that of fresh controls (see fresh control graph **430**) after rehydration was complete. Those directly rehydrated (see fresh control graph **430**) are swollen by many fold as indicated on the log scale of volume (fl) axis in Figure 43.

Referring now to Figures 44 and 45 for illustrating exemplarily how embodiments of the preservative solution effects membrane phase behavior, that membrane lipids in human platelets show a cooperative phase transition between about 10 and 20°C. Figure 44 is a graph illustrating the effects of prehydration (over water vapor) on phase behavior of freeze-dried platelets. More specifically shown in Figure 44 are scattered-point graphs **440** (directly rehydrated platelets scattered points), **442** (prehydrated platelets scattered points), and **444** (fresh control platelets scattered points). As broadly illustrated by scattered-point graph **440** in Figure 44, direct rehydration alters phase transition significantly (data obtained with Fourier transform infrared spectroscopy). The data represented by scattered-point graphs **440**, **442**, and **444** show clear differences in the phase transitions, with the prehydrated samples showing phase transitions essentially identical to fresh controls (see scattered-point graph **442** vs. scattered-point graph **444**). The directly rehydrated samples are clearly different as shown by scattered-point graph **440**.

Figure 45 is a graph illustrating the effects of prehydration (over water vapor) on the cooperativity of phase transition. More specifically shown in Figure 45 are fresh

control platelets curve **450** (a reference line for fully hydrated platelets that had never been dehydrated) and rehydrated platelets curve **452**. Fresh control platelets curve **450** and rehydrated platelets curve **452** illustrate that prehydration of the platelets over water vapor (see rehydrated platelets curve **452**) returned the phase transition parameter to nearly that of fresh control platelets ( see fresh control platelets curve **450**).

Figures 46 and 47 illustrate exemplarily how embodiments of the preservative solution effects protein secondary structure. Maintenance of protein secondary structure, and therefore function, is preferably required for platelet function after rehydration. Protein secondary structure was probed employing Fourier transform infrared spectroscopy, with the results shown in Figure 46 as a graph illustrating the effects of prehydration and/or direct rehydration on protein secondary structure in freeze-dried platelets, with direct rehydration altering protein secondary structure relative to controls. More specifically shown in Figure 46 are directly rehydrated curve **460**, prehydrated curve **464**, and fresh control platelet curve **462**. Figure 46 illustrates that protein secondary structure is essentially identical in fresh control platelets and in prehydrated platelets that were subsequently rehydrated (see fresh control platelet curve **462** vs. prehydrated curve **464**). Platelets that were directly rehydrated, as depicted by directly rehydrated curve **460**, show clear changes in the spectrum (i.e., the absorbance(relative units)), indicating damage to the protein secondary structure.

Figure 47 is a graph illustrating the effects of prehydration and/or direct rehydration on protein secondary structure in freeze-dried platelets, with prehydration returning protein secondary structure to control levels before direct



rehydration in bulk water. More specifically shown in Figure 47 are directly rehydrated platelet curve **470**, and fresh control platelet curve **474**. Figure 47 illustrates the recorded changes in one type of structure ( $\beta$ -sheet) as a function of water content, as was done with respect to the membrane protein structure. As shown by rehydrated platelet curve **470**, prehydration returned the protein secondary structure to that seen in fully hydrated platelets, as represented by fresh control platelets curve **474**. Thus, protein secondary structure in platelets prehydrated to 0.3 g H<sub>2</sub>O/g dry wt returned to fresh control levels before the platelets which were directly rehydrated in liquid water.

As was seen for the freeze-dried platelets, the freeze-dried eukaryotic cells, whether by themselves, as a component of a vial-compatible structure or matrix, may be packaged so as to prevent rehydration until desired. As previously indicated for platelets, the packaging may be any of the various suitable packaging for therapeutic purposes, such as made from foil metallized plastic materials, and moisture barrier plastics (e.g. high-density polyethylene or plastic films that have been created with materials such as SiO<sub>x</sub>), cooling the preservative (trehalose) loaded eukaryotic cells to below their freezing point, and lyophilizing the cooled eukaryotic cells. The trehalose loading preferably includes incubating the eukaryotic cells at a temperature from greater than about 25°C to less than about 50°C with a trehalose solution having up to about 50 mM trehalose therein. The process of using such a dehydrated cell composition comprises rehydrating the eukaryotic cells, which may be with any suitable aqueous solution, such as water. As previously indicated for platelets, the rehydration preferably includes a prehydration step sufficient to bring the water

content of the freeze-dried eukaryotic cells to between 35 weight percent to about 50 weight percent.

When reconstitution is desired, prehydration of the freeze-dried eukaryotic cells in moisture saturated air followed by rehydration is preferred. Use of prehydration yields eukaryotic cells with much more dense appearance and with no balloon eukaryotic cells being present. Prehydrated previously lyophilized eukaryotic cells resemble fresh eukaryotic cells after rehydration. This is illustrated, for example, by Figs. 16C, 17A and 17B. As can be seen in these figures, previously freeze-dried eukaryotic cells can be restored to a viable condition having an appearance of fresh eukaryotic cells.

Prehydration is preferably conducted in moisture saturated air, most preferably prehydration is conducted at about 37°C for about one hour to about three hours. The preferred prehydration step brings the water content of the freeze-dried eukaryotic cells to between about 20 weight percent to about 50 weight percent. The prehydrated eukaryotic cells may then be fully rehydrated. Rehydration may be with any aqueous based solutions (e.g., water), depending upon the intended application.

In additional embodiments of the present invention, it has been further discovered that many of the general findings with respect to platelets and eukaryotic cells are broadly applicable to erythrocytic cells. The term "erythrocytic cell" is used to mean any red blood cell. Mammalian, particularly human, erythrocytes are preferred. Suitable mammalian species for providing erythrocytic cells include by way of example only, not only human, but also equine, canine, feline, or endangered species.

The erythrocytic cells preferably contain an alcohol, more preferably an alcohol in a concentration ranging from about 10 wt. % to about 50 wt. %. In a preferred embodiment of the invention, the alcohol comprises a sterol, preferably a steroid alcohol containing the common steroid nucleus, plus an 8 to 10-carbon-atom side-chain and a hydroxyl group. It is known that sterols are widely distributed in plants and animals, both in the free form and esterified to fatty acids. Preferably, the steroid alcohol contained in the erythrocytic cells comprises cholesterol (cholesterin: 5-cholesten-3- $\beta$ -ol),  $C_{27}H_{45}OH$ , in a concentration ranging from about 10 wt. % to about 50 wt. %.

Cholesterol is an important mammalian (i.e., animal) sterol. Cholesterol is also the most common animal sterol, a monohydric secondary alcohol of the cyclopentenophenanthrene (4-ring fused) system, containing one double bond. It occurs in part as the free sterol and in part esterified with higher fatty acids as a lipid in human blood serum. The primary precursor in biosynthesis appears to be acetic acid or sodium acetate. It is known that cholesterol in the mammalian system is the precursor of bile acids, steroid hormones, and provitamin D3.

In a preferred embodiment of the invention, the erythrocytic cells comprise from about 10 wt. % to about 50 wt. % cholesterol having the following formula:

Thus, further compositions and further embodiments of the present invention include erythrocytic cells that have been manipulated (e.g., by freeze-drying) or modified (e.g., loaded with preservatives) and that are useful for well known therapeutic applications. We have discovered that alcohol-containing erythrocyte cells include alcohol-containing erythrocyte membranes, and have two phase transition temperature ranges, more specifically two weakly cooperative phase

transition temperature ranges respectively having a temperature range ranging from about 7°C to about 21°C (e.g., about 14.4°C  $\pm$  1.3°C) and from about 25°C to about 44°C (e.g., about 34.2°C  $\pm$  1.4°C). We have also discovered that removing at least part of the alcohol (e.g., a steroid alcohol, such as cholesterol) from the erythrocyte cells including the erythrocyte membranes results in an increase in the cooperativity of the two phase transition temperature ranges, as well as a formation of a third or intermediate phase temperature range. More specifically, we have discovered that after removal of at least about 10% by wt. of cholesterol, more preferably at least about 30% by wt. of cholesterol, alcohol reduced (i.e., sterol reduced) erythrocytic cells are produced preferably having from about 20% by weight to about 40% by weight alcohol, more preferably from about 20% by weight to about 30% by weight alcohol (e.g., cholesterol).

The alcohol or sterol reduced erythrocytic cells have a first or low phase transition temperature range greater than about 2°C, an intermediate phase temperature range greater than about 20°C, and a high phase transition temperature range greater than about 30°C. Preferably, the low phase transition temperature range ranges from a temperature greater than about 2°C to a temperature less than or equal to about 20°C (e.g., from about 12°C to about 18°C, such as about 15.3°C  $\pm$  about 0.8°C), the intermediate phase transition temperature range ranges from a temperature greater than about 20°C to a temperature less than or equal to about 30°C (e.g., from about 23°C to about 29°C, such as about 26.0°C  $\pm$  about 0.8°C), and the high phase transition 20 temperature range ranges from a temperature greater than about 30°C to a temperature less than or equal to about 50°C (e.g., from about 30°C to about 40°C, or

from about 32°C to about 38°C, such as about 35.4°C  $\pm$  about 1.5°C).

Our discovery of at least three phase transition temperature ranges, including increasing the cooperativity of the phase transitions, for the alcohol-reduced erythrocytic cells and erythrocytic membranes suggests improving the preservation of erythrocytic cells by optimizing loading erythrocytic cells with a preservative (e.g., an oligosaccharide, such as trehalose), and by optimizing the storage and rehydration of erythrocytic cells. Fundamental knowledge about membrane phase (i.e., phospholipid) transition temperatures is of practical importance for determining the in vitro storage conditions of erythrocytes in blood banks. We anticipate that preferably a temperature within the high phase transition temperature range (e.g., a temperature about 34°C) can be used to load erythrocytes with the oligosaccharide, such as trehalose, or any other lyoprotectant, and that intracellular preservative (e.g., a oligosaccharide including trehalose) allows the erythrocytes to survive freeze-drying. Freeze-dried erythrocytic cells will find broad applications in the field of medicine, pharmaceuticals, and biotechnology.

As was previously indicated for eukaryotic cells, one of the salient components for compositions and apparatus of additional embodiments of the present invention, when cell preservation will be assisted by freeze-drying, is the oligosaccharide, preferably trehalose, because we have discovered that alcohol-reduced erythrocytic cells which are effectively loaded with trehalose preserve biological properties during freeze drying (and rehydration). This preservation of biological properties, such as the immediate restoration of viability following rehydration, is necessary so that the

erythrocytic cells following preservation can be successfully used in a variety of well known therapeutic applications. Preferably, the preparation of preserved erythrocytic cells in accordance with embodiments of the present invention broadly comprises the steps of providing a source of erythrocytic cells having an alcohol (e.g., a steroid alcohol such as cholesterol), removing at least a portion of the alcohol from the erythrocytic cells, loading the erythrocytic cells with a protective preservative (e.g., an oligosaccharide), cooling the loaded erythrocytic cells to below  $-32^{\circ}\text{C}$ , and lyophilizing the cooled erythrocytic cells. In a preferred embodiment of the invention, the alcohol-reduced erythrocytic cells including erythrocytic membranes, comprise the three phase transition temperature ranges, and are loaded with the protective preservative at a temperature within one of the three phase transition temperature ranges. In erythrocytes from some species, it may not be necessary to remove any alcohol.

As was previously mentioned for eukaryotic cells, the source of the erythrocytic cells may be any suitable source. Regardless of the source of the erythrocytic cells, obtained erythrocytic cells typically comprise an alcohol, more preferably sterol, or a steroid alcohol such as cholesterol, at least a portion of which is to be removed to produce alcohol-reduced, more specifically, steroid/steroid alcohol-reduced or cholesterol-reduced erythrocytic cells having at least the three phase transition temperature ranges. The alcohol (e.g., cholesterol) may be removed from the erythrocytic cells/membranes by any suitable means or by any suitable manner. In a preferred embodiment of the invention, the alcohol is removed by incubating the erythrocytic cells, preferably incubating in an alcohol-removing medium containing an alcohol-

removing agent. More preferably, especially when the alcohol comprises cholesterol, the erythrocytic cells are incubated at a temperature ranging from about 25°C to about 50°C, more preferably from about 34°C to about 40°C, for a suitable time period (e.g., from 10 minutes to about three hours) in the presence of the cholesterol-removing medium containing a cholesterol-removing agent.

In a preferred embodiment of the invention, the cholesterol-removing medium comprises a buffer containing from about 1mM to about 10mM, preferably from about 1mM to about 5mM, of methyl- $\beta$ -cyclodextrin (M $\beta$ CD). The spirit and scope of the present invention includes not only M $\beta$ CD as the cholesterol-removing agent, but also any other cholesterol-removing agent for assisting in the removal of cholesterol from the erythrocytes including the erythrocytic membranes. In a preferred embodiment of the invention, after at least a portion of the cholesterol has been removed, the cholesterol-reduced erythrocytic cells/membranes preferably comprise from about 10 wt. % to about 50 wt. % cholesterol, more preferably from about 10 wt. to about 30 wt. % cholesterol, most preferably from about 20 wt. % to about 30 wt. %.

The alcohol-reduced erythrocytic cells are preferably loaded by incubating the alcohol-reduced erythrocytic cells in a buffer. The preservative (e.g., an oligosaccharide, such as trehalose) is preferably dissolved in the buffer, which includes any liquid solution capable of preserving living cells and tissue. Many types of buffers are known in the literature and available from any of the previously mentioned commercial suppliers for short term incubation of erythrocytes.

As was seen for nucleated cells, when the preservative to be loaded in the alcohol-reduced erythrocytic cells is

trehalose, the actual amount of trehalose dissolved in the buffer may vary, although considerations of the economical use of materials and labor, and considerations of the cryopreservation protocol, i.e., the choice of procedural steps used for cooling and thawing the alcohol-reduced erythrocytic cells together with the cooling and thawing rates, may affect the selection of concentration ranges that will provide the most efficient and effective preservation. In the case of trehalose for one embodiment of the present invention, the concentration of trehalose in the cryopreservation medium (i.e., the buffer plus added trehalose) ranges from about 10mM and about 1.5 M, preferably between about 100mM and about 500mM, in the cryopreservation medium. In another embodiment of the present invention, the concentration of trehalose in the cryopreservation medium ranges from about 10mM to less than about 100mM, such as from about 10mM to about 50mM, in the cryopreservation medium. The concentration of the alcohol-reduced erythrocytic cells in the cryopreservation medium that will provide optimal results may vary, and the concentration selected for use in any given procedure will be governed primarily by consideration of economy and efficiency. Effective results will generally be achieved with suspensions containing from about  $10^5$  to about  $10^{10}$  alcohol-reduced erythrocytic cells per milliliter of cryopreservation medium, preferably from about  $10^6$  to about  $10^9$  alcohol-reduced erythrocytic cells/mL, and most preferably from about  $10^7$  to about  $10^8$  alcohol-reduced erythrocytic cells/mL.

The amount of the preferred trehalose loaded inside the alcohol-reduced erythrocytic cells may be any suitable amount, preferably from about 10 mM to less than about 200 mM, more preferably from about 10 mM to about 150 mM, most preferably



from about 10 mM to about 100 mM, and is preferably achieved by incubating the alcohol-reduced erythrocytic cells to preserve biological properties during freeze-drying with a trehalose solution that has less than about 200 mM trehalose therein. As was found for platelets and eukaryotic cells, higher concentrations of trehalose during incubation are not preferred. The effective loading of trehalose is also accomplished by means of using a temperature that falls within one of the phase transition temperature ranges, preferably a temperature greater than about 30°C, more preferably a temperature ranging from about 30°C to less than about 50°C, most preferably from about 32°C to less than about 38°C, most preferably about 34°C. This is due to the discovery of the intermediate and high phase transition temperature ranges for alcohol-reduced erythrocytic cells/membranes. It is believed that the trehalose loading efficiency for alcohol-reduced erythrocytic cells increase at incubation temperatures equal to or above about 30°C up to about 50°C. Thus, it is believed that the Fig. 1 graph for platelets would be applicable for erythrocytic cells when the steep upwardly sloping line in Fig. 1 is extended to an incubation temperature of about 50°C.

The trehalose concentration in the exterior solution (that is, the loading buffer or cryopreservation medium) and the temperature during incubation together lead to a trehalose uptake that occurs primarily through defects that occur during the lipid phase transitions. Without being limited by theory, while we believe that pinocytosis is involved, it may be that the intermediate and/or high phase transition temperatures stimulate the pinocytosis. It is believed that other oligosaccharides when loaded in this intermediate and/or high phase transition temperatures in amounts analogous to trehalose

could have similar effects. Pinocytosed vesicles lyse over time which results in a homogeneous distribution of trehalose in the erythrocytic cells. It is also believed that the trehalose loading efficiency as a function of incubation time for erythrocytic cells would be comparable to that of platelets and eukaryotic cells. Thus, Fig. 2 would be representative of the trehalose loading efficiency as a function of incubation time for erythrocytic cells.

As was performed for lipid phase transitions in eukaryotic cells, lipid phase transitions in the alcohol-reduced erythrocytic cells are preferably measured by changes in membrane CH<sub>2</sub> vibrational frequency, using the Fourier transform infrared microscope coupled to an FTIR optical bench equipped with the temperature controller.

In preparing particularly preferred embodiments of the invention, alcohol-reduced erythrocytic cells/membranes may be loaded with trehalose by incubation at about 37°C for about twenty-four hours. The trehalose concentration in the loading buffer or cryopreservation medium is preferably about 35 mM, which results in an intracellular trehalose concentration of around 20 mM, but in any event is most preferably not greater than about 50 mM trehalose. At trehalose concentrations below about 50 mM, alcohol-reduced erythrocytic cells have a normal morphological appearance. After the alcohol-reduced erythrocytic cells have been effectively loaded with a preservative or protectant (e.g., an oligosaccharide, such as trehalose), then the loading buffer or cryopreservation medium is removed and the alcohol-reduced erythrocytic cells are contacted with a drying buffer (i.e., a freeze-drying buffer). Drying of alcohol-reduced erythrocytic cells after preservative loading may be carried out by suspending the alcohol-reduced erythrocytic cells in a

suitable drying solution containing a suitable water replacing molecule, such as trehalose and a bulking agent such as a salt, a starch, or an albumin. The drying buffer preferably also includes the preservative (e.g., trehalose), preferably in amounts up to about 200 mM, more preferably up to about 100 mM. Trehalose in the drying buffer assists in spatially separating the alcohol-reduced erythrocytic cells as well as stabilizing the alcohol-reduced erythrocytic membranes on the exterior. The drying buffer preferably also includes a bulking agent (to further separate the alcohol-reduced erythrocytic cells). As previously indicated for eukaryotic cells, albumin may serve as a bulking agent, but other polymers may be used with the same effect. Suitable other polymers, for example, are water-soluble polymers such as HES and dextran.

The preservative (trehalose) loaded alcohol-reduced erythrocytic cells in the drying buffer are then cooled to a temperature below about  $-32^{\circ}\text{C}$ . A cooling (i.e. freezing) rate is preferably between  $-30^{\circ}\text{C}$  and  $-1^{\circ}\text{C}/\text{min.}$ , and more preferably between about  $-2^{\circ}\text{C}/\text{min}$  to  $-5^{\circ}\text{C}/\text{min}$ . The lyophilization step is preferably conducted at a temperature below about  $-32^{\circ}\text{C}$ , for example conducted at about  $-40^{\circ}\text{C}$ .

In one embodiment of the present invention, drying may be continued until about 95 weight percent of water has been removed from the alcohol-reduced erythrocytic cells. During the initial stages of lyophilization, the pressure is preferably at about  $10 \times 10^{-6}$  Torr. As the cell samples dry, the temperature may be raised to be warmer than  $-32^{\circ}\text{C}$ . Based upon the bulk of the cell samples, the temperature, and the pressure, it may be empirically determined what the most efficient temperature values should be in order to maximize the evaporative water loss. For this embodiment of the invention, freeze-dried

alcohol-reduced erythrocytic cell compositions may have less than about 5 weight percent water.

In another embodiment of the invention, drying of the alcohol-reduced erythrocytic cells is continued until the water content of the alcohol-reduced erythrocytic cells is equal to or less than about 0.30 grams of water per gram of dry weight alcohol-reduced erythrocytic cells, more preferably equal to or less than about 0.20 grams of water per gram of dry weight alcohol-reduced erythrocytic cells. Preferably, the water content of the dried (e.g., freeze-dried) alcohol-reduced erythrocytic cells is maintained from about 0.00 gram, preferably from about 0.05 gram, of residual water per gram of dry weight alcohol-reduced erythrocytic cells to about 0.20 gram of residual water per gram of dry weight alcohol-reduced erythrocytic cells. For this embodiment of the invention, dehydration does not necessarily mean removal of 100% contained water. It has been discovered that by retention of at least about 0.3 gm of water per gm dry weight alcohol-reduced erythrocytic cells optimizes survival percentage of the alcohol-reduced erythrocytic cells after removal from the lyophilizer and rehydration is substantially increased.

As was seen for the freeze-dried platelets and freeze-dried eukaryotic cells, the freeze-dried alcohol-reduced erythrocytic cells, whether by themselves, as a component of a vial compatible structure or matrix, may be packaged so as to prevent rehydration until desired. As previously indicated for platelets and eukaryotic cells, the packaging may be any of the various suitable packaging for therapeutic purposes, such as made from foil metallized plastic materials, and moisture barrier plastics (e.g. high-density polyethylene or plastic films that have been created with materials such as SiO<sub>x</sub>), cooling the preservative

(trehalose) loaded alcohol-reduced erythrocytic cells to below their freezing point, and lyophilizing the cooled alcohol-reduced erythrocytic cells. As was further previously indicated for platelets and eukaryotic cells, the trehalose loading preferably includes incubating the alcohol-reduced erythrocytic cells at a temperature from greater than about 25°C to less than about 50°C with a trehalose solution having up to about 50 mM trehalose therein. The process of using such a dehydrated cell composition comprises rehydrating the alcohol-reduced erythrocytic cells, which may be with any suitable aqueous solution, such as water. As previously mentioned for platelets and eukaryotic cells, the rehydration preferably includes a prehydration step sufficient to bring the water content of the freeze-dried alcohol-reduced erythrocytic cells to between about 20 weight percent to about 50 weight percent, preferably between about 20 weight percent and about 40 weight percent.

When reconstitution is desired, prehydration of the freeze-dried alcohol-reduced erythrocytic cells in moisture saturated air followed by rehydration is preferred. Use of prehydration yields alcohol-reduced erythrocytic cells with much more dense appearance and with no balloon alcohol-reduced erythrocytic cells being present. Prehydrated previously lyophilized alcohol-reduced erythrocytic cells resemble fresh erythrocytic cells after rehydration. Prehydration is preferably conducted in moisture saturated air, most preferably prehydration is conducted at about 37°C for about one hour to about three hours. The preferred prehydration step brings the water content of the freeze-dried alcohol-reduced erythrocytic cells to between about 20 weight percent to about 40 weight percent. The prehydrated alcohol-reduced erythrocytic cells may then be fully rehydrated.

Rehydration may be with any aqueous based solutions (e.g., water), depending upon the intended application.

Embodiments of the present invention will be illustrated by the following set forth examples which are being given to set forth the presently known best mode and by way of illustration only and not by way of any limitation. All parameters such as concentrations, mixing proportions, temperatures, rates, compounds, etc., submitted in these examples are not to be construed to unduly limit the scope of the invention. Abbreviations used in the examples, and elsewhere, are as follows:

DMSO = dimethylsulfoxide

ADP = adenosine diphosphate

PGE1 = prostaglandin E1

HES = hydroxy ethyl starch

FTIR = Fourier transform infrared spectroscopy

EGTA = ethylene glycol-bis(2-aminoethyl ether) N,N,N',N',  
tetra-acetic acid

TES = N-tris (hydroxymethyl) methyl-2-aminoethane-sulfonic  
acid

HEPES = N-(2-hydroxyl ethyl) piperazine-N'-(2-  
ethanesulfonic acid)

PBS = phosphate buffered saline

HSA = human serum albumin

BSA = bovine serum albumin

ACD = citric acid, citrate, and dextrose

M $\beta$ CD = methyl- $\beta$ -cyclodextrin

## EXPERIMENTAL

### EXAMPLE 1

Washing of Platelets. Platelet concentrations were obtained from the Sacramento blood center or from volunteers in our laboratory. Platelet rich plasma was centrifuged for 8 minutes at 320 x g to remove erythrocytes and leukocytes. The supernatant was pelleted and washed two times (480 x g for 22 minutes, 480 x g for 15 minutes) in buffer A (100 mM NaCl, 10 mM KCl, 10 mM EGTA, 10 mM imidazole, pH 6.8). Platelet counts were obtained on a Coulter counter T890 (Coulter, Inc., Miami, Florida).

Loading of Lucifer Yellow CH into Platelets. A fluorescent dye, lucifer yellow CH (LYCH), was used as a marker for penetration of the membrane by a solute. Washed platelets in a concentration of  $1-2 \times 10^9$  platelets/ml were incubated at various temperatures in the presence of 1-20 mg/ml LYCH. Incubation temperatures and incubation times were chosen as indicated. After incubation the platelets suspensions were spun down for 20 s at 14,000 RPM (table centrifuge), resuspended in buffer A, spun down for 20 s in buffer A and resuspended. Platelet counts were obtained on a Coulter counter and the samples were pelleted (centrifugation for 45 s at 14,000 RPM, table centrifuge). The pellet was lysed in 0.1% Triton buffer (10 mM TES, 50 mM KCl, pH 6.8). The fluorescence of the lysate was measured on a Perkin-Elmer LSS spectrofluorimeter with excitation at 428 nm (SW 10 nm) and emission at 530 nm (SW 10 nm). Uptake was calculated for each sample as nanograms of LYCH per cell using a

standard curve of LYCH in lysate buffer. Standard curves of LYCH, were found to be linear up to 2000 run  $\text{ml}^{-1}$ .

Visualization of cell-associated Lucifer Yellow. LYCH loaded platelets were viewed on a fluorescence microscope (Zeiss) employing a fluorescein filter set for fluorescence microscopy. Platelets were studied either directly after incubation or after fixation with 1% paraformaldehyde in buffer. Fixed cells were settled on poly-L-lysine coated cover slides and mounted in glycerol.

Loading of Platelets with Trehalose. Washed platelets in a concentration of  $1-2 \times 10^9$  platelets/ml were incubated at various temperatures in the presence of 1-20 mg/ml trehalose. Incubation temperatures were chosen from 4°C to 37°C. Incubation times were varied from 0.5 to 4 hours. After incubation the platelet solutions were washed in buffer A two times (by centrifugation at 14,000 RPM for 20 s in a table centrifuge). Platelet counts were obtained on a coulter counter. Platelets were pelleted (45 S at 14,000 RPM) and sugars were extracted from the pellet using 80% methanol. The samples were heated for 30 minutes at 80°C. The methanol was 10 evaporated with nitrogen, and the samples were kept dry and redissolved in  $\text{H}_2\text{O}$  prior to analysis. The amount of trehalose in the platelets was quantified using the anthrone reaction (Umbreit et al., *Mamometric and Biochemical Techniques*, 5th Edition, 1972). Samples were redissolved in 3 ml  $\text{H}_2\text{O}$  and 6 ml anthrone reagents (2 g anthrone dissolved in 10M sulfuric acid). After vortex mixing, the samples were placed in a boiling water bath for 3 minutes. Then the samples were cooled on ice and the absorbance was measured at 620 nm on a Perkin Elmer spectrophotometer. The amount of platelet associated trehalose was determined using a standard curve of trehalose.



Standard curves of trehalose were found to be linear from 6 to 300  $\mu\text{g}$  trehalose per test tube.

Quantification of Trehalose and LYCH Concentration. Uptake was calculated for each sample as micrograms of trehalose or LYCH per platelet. The internal trehalose concentration was calculated assuming a platelet radius of 1.2  $\mu\text{m}$  and by assuming that 50% of the platelet volume is taken up by the cytosol (rest is membranes). The loading efficiency was determined from the cytosolic trehalose or LYCH concentration and the concentration in the loading buffer.

Fig. 1 shows the effect of temperature on the loading efficiency of trehalose into human platelets after a 4 hour incubation period with 50 mM external trehalose. The effect of the temperature on the trehalose uptake showed a similar trend as the LYCH uptake. The trehalose uptake is relatively low at temperatures of 22°C and below (below 5%), but at 37°C the loading efficiency of trehalose is 35% after 4 hours.

When the time course of trehalose uptake is studied at 37°C, a biphasic curve can be seen (Fig. 2). The trehalose uptake is initially slow ( $2.8 \times 10^{-11} \text{ mol/m}^2\text{s}$  from 0 to 2 hours), but after 2 hours a rapid linear uptake of  $3.3 \times 10^{-10} \text{ mol/m}^2\text{s}$  can be observed. The loading efficiency increases up to 61% after an incubation period of 4 hours. This high loading efficiency is a strong indication that the trehalose is homogeneously distributed in the platelets rather than located in pinocytosed vesicles.

The uptake of trehalose as a function of the external trehalose concentration is shown in Fig. 3. The uptake of trehalose is linear in the range from 0 to 30 mM external trehalose. The highest internal trehalose concentration is

obtained with 50 mM external trehalose. At higher concentrations than 50 mM the internal trehalose concentration decreases again. Even when the loading buffer at these high trehalose concentrations is corrected for isotonicity by adjusting the salt concentration, the loading efficiency remains low. Platelets become swollen after 4 hours incubation in 75 mM trehalose.

The stability of the platelets during a 4 hours incubation period was studied using microscopy and flow cytometric analysis. No morphological changes were observed after 4 hours incubation of platelets at 37°C in the presence of 25 mM external trehalose. Flow cytometric analysis of the platelets showed that the platelet population is very stable during 4 hours incubation. No signs of microvesicle formation could be observed after 4 hours incubation, as can be judged by the stable relative proportion of microvesicle gated cells (less than 3%). The formation of microvesicles is usually considered as the first sign of platelet activation (Owens et al., *Trans. Med. Rev.*, 8, 27-44, 1994). Characteristic antigens of platelet activation include: glycoprotein 53 (gp53, a lysosomal membrane marker), PECAM-1 (platelet endothelial cell adhesion molecule-1, an alpha granule constituent), and P-selection (an alpha granule membrane protein).

## EXAMPLE 2

Washing Platelets. Platelets were obtained from volunteers in our laboratory. Platelet rich plasma was centrifuged for 8 minutes at 320 x g to remove erythrocytes and leukocytes. The supernatant was pelleted and washed two times (480 x g for 22 minutes, 480 x g for 15 minutes) in buffer A (100 mM NaCl, 10 mM

KCl, 10 mM EGTA, 10 mM imidazole, 10  $\mu$ g/ml PGE1, pH 6.8). Platelet counts were obtained on a Coulter counter T890 (Coulter, Inc., Miami, Florida).

Loading Platelets with Trehalose. Platelets were loaded with trehalose as described in Example 1. Washed platelets in a concentration of  $1-2 \times 10^9$  platelets/ml were incubated at 37°C in buffer A with 35 mM trehalose added. Incubation times were typically 4 hours. The samples were gently stirred for 1 minute every hour. After incubation the platelet solutions were pelleted (25 sec in a microfuge) and resuspended in drying buffer (9.5 mM HEPES, 142.5 mM NaCl, 4.8 mM KCl, 1 mM  $MgCl_2$ , 30 mM Trehalose, 1% Human Serum Albumin, 10  $\mu$ g/ml PGE1). In the aggregation studies no PGE1 was added in the drying buffer. Trehalose was obtained from Pfahnstiehl. Human serum albumin was obtained from Sigma.

Freezing and Drying. Typically 0.5 ml platelet suspensions were transferred in 2 ml Nunc cryogenic vials and frozen in a Cryomed controlled freezing device. Vials were frozen from 22°C to -40°C with freezing rates between -30 and -1°C/min and more often between -5 and -2°C/min. The frozen solutions were transferred to a -80°C freezer and kept there for at least half an hour. Subsequently the frozen platelet suspensions were transferred in vacuum flasks that were attached to a Virtis lyophilizes. Immediately after the flasks were hooked up to the lyophilizer, they were placed in liquid nitrogen to keep the samples frozen until the vacuum returned to  $20 \times 10^{-6}$  Torr, after which the samples were allowed to warm to the sublimation temperature. The condenser temperature was -45°C. Under these conditions, sample temperature during primary drying is about -40°C, as measured with a thermocouple in the sample. It is

important to maintain the sample below  $T_g$  for the excipient during primary drying ( $-32^{\circ}\text{C}$  for trehalose).

Rehydration. Vials with originally 0.5 ml platelet suspension were rehydrated in 1 ml PBS buffer/water (1/1). PBS buffer was composed of 9.4 mM  $\text{Na}_2\text{HPO}_4$ , 0.6 mM  $\text{KH}_2\text{PO}_4$ , 100 mM NaCl, pH7.2). In a few experiments PGE1 was added to the rehydration buffer in a condition of 10  $\mu\text{g/ml}$  or rehydration was performed in plasma/water (1 / 1).

Prehydration. Platelet lyophilisates were prehydrated in a closed box with moisture saturated air at  $37^{\circ}\text{C}$ . Prehydration times were between 0 and 3 hours.

Recovery. The numerical recovery of lyophilized and (p)rehydrated platelets was determined by comparing the cell count with a Coulter count T890 (Coulter, Inc., Miami, Florida) before drying and after rehydration. The morphology of the rehydrated platelets was studied using a light microscope. For this purpose platelets were fixed in 2% paraformaldehyde or glutaraldehyde and allowed to settle on poly-L-lysine coated coverslides for at least 45 minutes. After this the coverslides were mounted and inspected under the microscope. The optical density of freeze-dried and rehydrated platelets was determined by measuring the absorbance of a platelet suspension of  $1.0 \times 10^8$  cells/ml at 550 nm on a spectrophotometer.

Aggregation studies. Dried platelets were rehydrated (after 2 hour prehydration) with 2 aliquots of platelet free plasma (plasma was centrifuged for 5 minutes at  $3800 \times g$ ) diluted with water in 1/1 ratio. Half ml aliquots of this platelet suspension were transferred to aggregation cuvettes with a magnetic stirrer. The response of the platelets to thrombin was tested by adding thrombin (1 U/ml) to the platelet suspension at  $37^{\circ}\text{C}$

under stirring conditions. After 3 minutes thrombin treated platelet suspensions were inspected for clots and cell counts were done on a Coulter Counter T890.

Direct rehydration tends toward cell lysis and prehydration leads to aggregation when the cell concentration is  $10^9$  cells/ml in the drying buffer. We found also that recovery of prehydrated and rehydrated platelets depends on the cell concentration in the drying buffer. The recovery drops to very low values if the cell concentration is higher than  $3 \times 10^8$  cells/ml. At concentrations below  $3 \times 10^8$  cells/ml, the recovery is around 70%, and no aggregates were visible. Prehydration resulted in denser cells and the absence of balloon cells.

Longer prehydration times than 90 minutes did not further improve the cellular density, but slightly activated the platelets. The water content of the pellet increases with increasing prehydration time, and preferably is between about 35% and 50% at the moment of rehydration.

At higher water contents than 50% water droplets become visible in the lyophilisate (which means that the platelets are in a very hypertonic solution).

---

As described by Example 1, platelets were loaded with trehalose by incubation at 37°C for 4 hours in buffer A with 35 mM trehalose, which yielded platelets with intracellular trehalose concentration of 15-25 mM. After incubation, the platelets were transferred to drying buffer with 30 mM trehalose and 1% HSA as the main excipients.

The directly rehydrated platelets had a high numerical recovery of 85%, but a considerable fraction (25-50%) of the cells was partly lysed and had the shape of a balloon. Directly

rehydrated platelets were overall less dense when compared with fresh platelets.

The numerical recovery of platelets that were prehydrated in moisture saturated air was only 25% when the platelet concentration was  $1 \times 10^9$  cells/ml in the drying buffer. This low recovery was due to aggregates that were formed during the prehydration period. But the cells that were not aggregated were more dense than the directly rehydrated platelets and resembled that of fresh platelets.

Since it appears desirable to dilute the platelets to prevent aggregation during the prehydration step, it may be necessary for clinical applications to concentrate the platelets following rehydration. We therefore also tested the stability of the rehydrated platelets with respect to centrifugation and found that the directly rehydrated platelets had 50% recovery after centrifugation, while the prehydrated ones had 75% recovery following centrifugation. Thus, we conclude that the inventive platelets can be concentrated without ill effect.

### EXAMPLE 3

We view trehalose as the main lyoprotectant in the drying buffer. However, other components in the drying buffer, such as albumin, can improve the recovery. In the absence of external trehalose in drying buffer, the numerical recovery is only 35%. With 30 mM trehalose in the drying buffer the recovery is around 65%. A combination of 30 mM trehalose and 1% albumin gave a numerical recovery of 85%.

### EXAMPLE 4

Typically 0.5 ml platelet suspensions were transferred in 2 ml Nunc cryogenic vials and frozen in a Cryomed controlled freezing device. Vials were frozen from 22°C to -40°C with freezing rates between -30°C/min and -1°C/min and more often between -5°C and -2°C/min. The frozen solutions were transferred to a -80°C freezer and kept there for at least half an hour. Subsequently the frozen platelet suspensions were transferred in vacuum flasks that were attached to a Virtus lyophilizer. Immediately after the flasks were hooked up to the lyophilizer, they were placed in liquid nitrogen to keep the samples frozen until the vacuum returned to  $20 \times 10^{-6}$  Torr, after which the samples were allowed to warm to the sublimation temperature. The condensor temperature was -45°C. Under these conditions, sample temperature during primary drying is about -40°C, as measured with a thermocouple in the sample. It is important to maintain the sample below  $T_g$  for the excipient during primary drying (-32°C for trehalose). Only minor differences in recovery were found as a function of the freezing rate. The optimal freezing rate was found to be between 2°C and 5°C/minute.

#### EXAMPLE 5

Response of freeze-dried platelets to thrombin (1 U/ml) was compared with that of fresh platelets. The platelet concentration was  $0.5 \times 10^8$  cells/ml in both samples. 500  $\mu$ l platelets solution was transferred into aggregation vials. Thrombin was added to the samples and the samples were stirred for 3 minutes at 37°C. The cell counts that were determined after 3 minutes were 0 for both the fresh and the freeze-dried platelets. The response to thrombin was determined by a cleavage in glycoprotein Ib-(GPIb). This was detected by using monoclonal antibodies and flow cytometry. Thus, the

pattern seen after addition of thrombin was a reduced amount of GP 1b on the platelet surface.

The response of lyophilized, prehydrated, and rehydrated platelets (Examples 1 and 2) to thrombin (1 U/ml) was found to be identical compared with that of fresh platelets. In both fresh and rehydrated platelets a clot was formed within 3 minutes at 37°C. These clots are illustrated by Fig. 8, panels (A) and (B). When cell counts were done with the Coulter counter, we found no cells present, indicating that all platelets participated in forming the clot illustrated in panel (B).

#### EXAMPLE 6

Reactions with other agonists were studied. Platelet suspensions of the inventive platelets were prepared with  $50 \times 10^6$  platelets/ml. Different agonists were then added and subsequently counted with a Coulter counter to determine the percentage of platelets involved in the visually observable clot formation. The cell count was between 0 and  $2 \times 10^6$  platelets/ml: after 5 minutes with 20 µg/ml collagen; after 5 minutes with 20 µM ADP; after 5 minutes with 1.5 mg/ml ristocetin. This means that the percentage of platelets that are involved in clot formation is between 95-100% for all the agonists tested. The agonist concentrations that were used are all physiological. In all cases the percentage of clotted platelets was the same as fresh control platelets.

#### EXAMPLE 7

The procedures performed in this example were for mesenchymal stem cells, and illustrate cell culture, lipid



phase transitions, cell loading, freeze-drying, rehydration and membrane phase transition.

Cell Culture. Mesenchymal stem cells (MSCs) supplied by Osiris Therapeutics were grown with Dulbecco's Modified Eagle's Medium (D-MEM) supplemented with 10% v/v fetal bovine serum (FBS) in T-185 Culture Flasks (Nalge-Nunc). Serum-supplemented cells were incubated at 37°C and 5% CO<sub>2</sub>.

Fourier Transform Infrared Spectroscopy. MSCs harvested by trypsinization were resuspended in 2 mL fresh medium, and the cells were allowed to settle for 30 min. The cell pellet was applied as a thin film between two CaF<sub>2</sub> windows and scanned by Fourier transform infrared (FTIR) spectroscopy on a Perkin Elmer Spectrum 2000. Data were collected from 3600 to 900 cm<sup>-1</sup> every 2°C between -7 and 50°C using a ramp rate of 2°C/min. Temperature was controlled by a Peltier device and monitored with a thermocouple attached directly to the sample windows.

Lucifer Yellow CH-Loading. MSCs were harvested by trypsinization, washed once and resuspended in fresh medium at a concentration of  $5.7 \times 10^6$  cells/mL. Lucifer yellow CH (LYCH) was added to a concentration of 10.6 mM, and cells were tumbled in a flask at 37°C for 3.5 hours. Aliquots of cells were removed at several time points and washed twice with DPBS. The pellet was split between two treatments. The fluorescence intensity of the cells was measured with a Perkin Elmer LS 50B luminescence spectrometer, using an excitation wavelength of 428 nm and an emission wavelength of 530 nm. In addition, cells from each time point were fixed in 1 % paraformaldehyde, mounted on poly-L-lysine coated coverslips, and photographed with a Zeiss inverted fluorescent microscope, model ICM 405.

Freeze-Drying Flask Preparation. Freeze-drying flasks were prepared using Nalge-Nunc T-25 flasks modified for this purpose. These flasks have 0.22  $\mu\text{m}$  filters to allow vapor transport without compromising sterility, and includes a thermocouple port to allow direct temperature measurement of the sample. Prior to freeze drying, the flasks were immersed in 70% ethanol to sterilize them after they were completely assembled. The flasks were then allowed to dry in a laminar flow hood.

Freeze-Drying. MSCs were initially loaded with trehalose by incubating them in medium supplemented with 90 mM trehalose for 24 hours. The cells were then harvested, washed and resuspended in freeze-drying buffer (130mM NaCl, 10mM HEPES (pH 7.2), 5mM KCl, 150mM trehalose, and 5.7% BSA (w/v)) to a final concentration of  $0.5 \times 10^6$  cells/mL. This cell suspension was added in 2.5 mL aliquots to freeze-drying flasks and transferred to the Lyostar lyophilizer. The samples were frozen first at  $5^\circ\text{C}/\text{min}$  to  $0^\circ\text{C}$ , then at  $2^\circ\text{C}/\text{min}$  to  $-60^\circ\text{C}$ . Once freeze-drying began, cells were maintained under vacuum at  $-30^\circ\text{C}$  for 180 minutes, then at  $-25^\circ\text{C}$  for 180 minutes. Finally, the cells were slowly ramped to room temperature over a 12-hour period under vacuum. With this protocol, the cells are freeze-dried in suspension, rather than as an attached culture.

Rehydration. Freeze-dried cells were rehydrated with a 1:3 mixture of  $\text{H}_2\text{O}$  (equal to the original volume dried) and growth medium containing fetal bovine serum. This rehydration solution was either added directly to the lyophilizate or following a 45-min "prehydration" at  $37^\circ\text{C}$  and 100% relative humidity. Micrographs were taken on a Zeiss inverted microscope using phase contrast or fluorescence modes using Kodak Ektachrome ASA 400 film.

Membrane Phase Transition. The membrane phase transition of hydrated MSCs was determined using FTIR spectroscopy, and Fig. 10 shows data sets for two independent experiments. The data points indicate the symmetric CH<sub>2</sub> stretching band position for each temperature, and the solid line shows the first derivative for one data set. Thus, Figure 10 is more specifically a graph illustrating temperatures for membrane phase transition in hydrated mesenchymal stem cells by Fourier transform infrared (FTIR) spectroscopy, with the solid line graph indicating the first derivative of the set of data shown in filled circles. The peaks in the first derivative indicate the steepest regions in the band position vs. temperature plots that correspond to membrane phase transition temperatures. Two main transitions are evident at approximately 15 and 35°C, a pattern which has been observed in other cell types as well. This information enables characterization of the physical nature of the MSC membrane. The relationship between the phase transition in the hydrated and dry states (+/- trehalose) provides important information regarding the necessity and length of the prehydration protocol.

#### EXAMPLE 8

The procedures performed in this example were also for mesenchymal stem cells, and illustrate cell loading, cell growth, and freeze-drying.

Lucifer Yellow-Loading. Mesenchymal stem cells were tested for their ability to take up solutes from the extracellular environment. The dye Lucifer yellow CH (LYCH) was used as a marker for this type of uptake as it is easily monitored, both by fluorescence spectroscopy and fluorescence microscopy. Figure 11 is a graph representing LYCH loading of mesenchymal stem

cells as monitored fluorescence spectroscopy (filled circles points) and viability as monitored trypan blue exclusion (filled squares points). The open symbols in Fig. 11 show fluorescence and viability data for control cells (no LYCH). Fig. 11 shows the progressive uptake of LYCH over a period of 3.5 hours as well as the viability (-70%), which was monitored in parallel by trypan blue exclusion. It is believed that -70% viability was due to a period of approximately 2.5 hours that the cells were at room temperature after being trypsinized but before the loading experiment began. It is believed that by proceeding immediately from trypsinization to the next step (i.e., the loading step) in the protocol, the viability improves.

Micrographs taken in phase contrast and fluorescence modes of LYCH-loaded cells are shown in Figs. 12A-12J. Figures 12A-12B are micrographs of the human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope 30 minutes following LYCH-loading, with Fig. 12A showing phase contrast images and all cells intact and Fig. 12B showing fluorescent images for the same cells of Fig. 12A and the LYCH uptake after 30 minutes. Figures 12C-12D are micrographs of the human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope 1 hour following LYCH-loading, with Fig. 12C showing phase contrast images and all cells intact and Fig. 12D showing fluorescent images for the same cells of Fig. 12C and the LYCH uptake after 1 hour. Figures 12E-12F are micrographs of the human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope 2 hours following LYCH-loading, with Fig. 12E showing phase contrast images and all cells intact and Fig. 12F showing fluorescent images for the same cells of Fig. 12E and the LYCH uptake after 2 hours. Figures 12G-12H are micrographs of the human mesenchymal stem cells taken at 630X on a Zeiss inverted

microscope 3.5 hours following LYCH-loading, with Fig. 12G showing phase contrast images and all cells intact and Fig. 12H showing fluorescent images for the same cells of Fig. 12G and the LYCH uptake after 3.5 hours. Figures 12I-12J are micrographs of a control sample (cells incubated in the absence of LYCH) of the human mesenchymal stem cells taken at 630X on a Zeiss inverted microscope and having no LYCH-loading of the cells, with Fig. 12I showing phase contrast images and all cells intact and Fig. 12J showing no fluorescent images for the same cells of Fig. 12I because the fluorescence is specific to LYCH and does not correspond to auto-fluorescence from the human mesenchymal stem cells.

Phase contrast images showed that all cells were intact. The fluorescence micrographs showed the progression of LYCH uptake over time. At earlier time points, the cytoplasm was only dimly stained, and bright punctate staining near the plasma membrane indicated dye uptake into vesicles. This suggests that the loading likely occurred via an endocytotic mechanism. At later time points, the cytoplasm was more brightly and uniformly stained, indicating that leakage from the vesicles raised the concentration of dye throughout the cells.

Growth Curves. MSCs were plated into 12-welled plates at approximately the same seeding density used for T-185 flasks in standard Osiris protocols (5900 cells/cm<sup>2</sup> with fluid volume of 0.189 mL/cm<sup>2</sup>). Three wells for each condition at each time point were trypsinized and counted. Data for cells grown in the presence of trehalose were lost for the first two time points. Figure 13 is a graph illustrating growth curves for the mesenchymal stem cells in the presence or absence of 90 mM trehalose with the open triangle data representing cells grown in standard medium for 24 hours, after which 90mM trehalose was

added. It is clear from Fig. 13 that trehalose did not interfere with growth of the cells up to the third day. Subsequently, the cell count started to drop significantly in the presence of trehalose, and thus, incubation of MSCs for more than two days in trehalose should be avoided.

Freeze-Drying Mesenchymal Stem Cells. Human MSCs were prepared for freeze-drying by a 24-hour incubation at 37°C in their standard growth medium plus 90 mM trehalose. The cells are likely to take up trehalose in a manner similar to that shown above for LYCH, as has been seen with platelets and epithelial 293H cells. Following the trehalose-incubation, MSCs were harvested, transferred to a freeze-drying buffer, and placed into two T-25 flasks modified for freeze-drying. The cell samples were freeze-dried on a Lyostar lyophilizer and rehydrated as detailed above. The freeze-dried cake was homogeneous and robust with no indications of collapse. The cells survived for several days following rehydration, as their plasma membranes were intact and their nuclei were clearly seen. In addition, some cells attached to the substrate and appeared to be initiating the stretched and spreading morphology. Overall health appeared better in the cell sample which had received the prehydration treatment prior to full rehydration.

Figure 14A is a micrograph at a 100X magnification of the healthy mesenchymal stem cell culture prior to harvest by trypsinization. Figure 14B is a micrograph at a 320X magnification of the healthy mesenchymal stem cell culture of Fig. 14A prior to harvest by trypsinization. Figure 15A is a 100X magnified image of the dry lyophilization "cake" of mesenchymal stem cells encased in strands of matrix containing trehalose and BSA. Figure 15B is a 100X magnified image of the prehydrated lyophilization "cake" of mesenchymal stem cells

encased in strands of matrix containing trehalose and BSA. Figure 16A is a micrograph of the mesenchymal stem cells magnified 100X following freeze-drying and rehydration. Figure 16B is a micrograph of the mesenchymal stem cells magnified 400X following freeze-drying and rehydration. Figure 16C is a micrograph of the mesenchymal stem cells magnified 400X following freeze-drying, initial prehydration, and rehydration. Figure 17A is a micrograph of the mesenchymal stem cells from the prehydrated sample at two days post rehydration, illustrating the attached cell and the beginning appearance of characteristic stretched morphology. Figure 17B is a micrograph of the mesenchymal stem cells from the prehydrated sample at five days post rehydration, with nuclei clearly visible in several of the cells.

#### EXAMPLE 9

The procedures performed in this example were for epithelial 293H cells, and illustrate cell loading, freeze-drying, prehydration, FTIR analysis, and rehydration.

Trehalose Loading. Epithelial 293H cells chosen to be loaded with trehalose were taken from a stock culture, trypsinized, washed, and seeded into a new T-75 flask containing normal growth medium with the addition of 90mM trehalose. The osmolarity of the medium was not adjusted, yielding a final culture medium osmolarity with trehalose of approximately 390 mOsm. Cells were allowed to grow in this state under normal incubation conditions for 72 hours. They were then harvested using standard protocols and resuspended in freeze-drying buffer immediately prior to the freeze-drying procedure. The freeze-drying buffer contained 130 mM NaCl, 10 mM HEPES (Na), 5mM KCl,

150 mM trehalose, and 14.2 g BSA (5.7%) w/v. The buffer was at pH 7.2 and was maintained at 37°C.

Freeze-dry. Freeze-drying protocols were developed to optimize drying using the T-25 Lyoflasks. Cells were initially frozen at 5°C/min to 0°C then at 2°C/min to -60°C. Once freeze-drying begins, cells were maintained under vacuum at -30°C for 180 minutes, then at -25°C for 180 minutes. Last, the cells are slowly ramped to room temperature over a 12 hour period under vacuum.

Figure 18A is a micrograph at 100X magnification of the epithelial 293H cells freeze-dried in trehalose, with the cells remaining whole and round, closely resembling their native hydrated state. Figure 18B is an enlarged view of the dashed square cell field in Fig. 18A with the arrows identifying exceptionally preserved cells. Figure 19A is a micrograph at 400X magnification of the epithelial 293H cells freeze-dried in trehalose, and showing two epithelial 293H cells imbedded within a freeze-drying matrix composed of trehalose, albumin, and salts, with the cells appearing whole, round, and completely engulfed within the matrix. Figure 19B is an enlarged view of the dashed square cell field in Fig. 19A with two epithelial cells respectively identified by an arrow.

Rehydration. Freeze-dried cells were either rehydrated directly with a rehydration buffer of 1:3 H<sub>2</sub>O to growth medium mixture, or were first prehydrated at 100% relative humidity for 45 min and then were fully rehydrated with the same rehydration buffer. Images were taken on a Zeiss inverted microscope using bright field or phase contrast at 100X, 320X, and 400X on Kodak Ektachrome ASA 400 film.



Figure 20A is a micrograph at 100X magnification of the epithelial 293H cells after 30 prehydration (45 min @ 100% relative humidity) and rehydration (1:3 ratio of H<sub>2</sub>O:growth medium), and showing a high number of intact, refractile cells. Figure 20B is an enlarged view of the dashed square cell field in Fig. 20A. Figure 21A is a micrograph at 320X magnification of the epithelial 293H cells 24 hours following rehydration, with refractile whole cells still visible. Figure 21B is an enlarged view of the dashed square cell field in Fig. 21A with a refractile cell marked by an arrow.

FTIR Analysis. The protocol used for analysis of membrane phase transitions by Fourier transform infrared spectroscopy (Perkin-Elmer Spectrum 2000) was as follows: Cells, either hydrated or dry, with or without trehalose, were placed between CaF<sub>2</sub> windows. These samples were scanned between 3600 and 900 cm<sup>-1</sup> over a range of temperatures with a ramping rate of 2°C/min. Raw spectra were then analyzed for changes in wavenumber of the symmetric CH<sub>2</sub> stretching vibration of membrane lipids (around 2850). Band position was graphed as a function of temperature, and first derivative analysis indicates the membrane phase transition temperatures. Dried samples were prepared by freeze-drying and were loaded onto the windows in a dry box.

Use of the BioDRI Flask for Freeze-Drying Epithelial 293H Cells. Following the freeze-drying procedure, the lyophilized epithelial 293H cells appeared to be optimally freeze-dried. The lyophilizate formed a dry cake that is indicative of proper drying, without having melted or collapsed. In the dry state, the cells appeared to be highly well-preserved in the trehalose/buffer matrix. Cells remained intact and round, similar to the shape and size seen in trypsinized epithelial 293H cells (see Figs. 18A, 18B, 19A and 19B). By maintaining the

cells' native structure, it appeared that the dried state encased within trehalose was sufficient.

Following rehydration, under both direct and prehydrated conditions, cells were mostly whole and intact following the addition of the rehydration buffer. Cells that were first prehydrated appeared more refractile than in the directly hydrated samples (see Fig. 20A and 20B). In both cases very few cells appeared lysed due to the reintroduction of water. Furthermore, cellular debris was almost completely absent from either condition. Overall, in the prehydrated sample, approximately 10% of the cells imaged appeared highly refractile initially (see Figs. 20A and 20B).

Twenty-four (24) hours following rehydration, the rehydrated epithelial 293H cells in culture were again observed. Those cells in the prehydrated condition appeared to be more refractile and attached more strongly to the growth surface than those in the non-prehydrated sample (see Figs. 21A and 21B). In the prehydrated condition, approximately 6 to 7% of the cells remained phase bright. It was apparent, however, that cellular debris became abundant and that many cells had lysed.

#### EXAMPLE 10

The physical membrane properties of human erythrocytes were studied *in situ* using Fourier transform infrared spectroscopy (FTIR). Erythrocyte membranes were found to have weakly cooperative phase transitions at 14°C and at 34°C. Cholesterol depletion by methyl- $\beta$ -cyclodextrin (M $\beta$ CD) resulted in a large increase in the cooperativity of these transitions, and led to the appearance of another phospholipid transition at 25°C. Multiple, sharp membrane phase transitions were observed after

five days cold storage (4°C), which indicated phase separation of the membrane lipids. Using fluorescence microscopy, it was determined that the lipid probe 1,1'-dioctadecyl-3,3,3',3'-tetramethyl-indocarbocyanine perchlorate (dil-C<sub>18</sub>) remained homogeneously distributed in the erythrocyte membrane during cold storage, suggesting that lipid domains were below the resolution limit of the microscope. Using thin layer chromatography, changes in the membrane lipid composition were detected during cold storage. By contrast, assessment of the amide-II band with FTIR showed that the overall protein secondary structure of hemoglobin was stable during cold stage.

Fundamental knowledge about membrane phase transition temperatures is of practical importance for determining the *in vitro* storage conditions of erythrocytes in blood banks. We anticipate that especially the phase transition temperature at around 34°C may be used to load erythrocytes with trehalose, or any other lyoprotectant and that intracellular trehalose allows the erythrocytes to survive freeze-drying. Freeze-dried erythrocytic cells will find broad applications in the field of medicine, pharmaceuticals and biotechnology.

### Experimental Procedures

Isolation of Erythrocytes Cholesterol Depletion Preparation of Ghosts, and Storage Conditions. Venous blood was collected from healthy adults, with informed consent, according to institutional protocols. Blood was anticoagulated with ACD (citric acid, citrate, dextrose). Whole blood was centrifuged at 320 x g for 8 minutes. The cellular pellet (red blood cells) was washed three times in PBS buffer (100 mM NaCl, 9.4 mM Na<sub>2</sub>HPO<sub>4</sub>, 0.6 mM KH<sub>2</sub>PO<sub>4</sub>, pH 7.4) prior to further analysis. Cells were

incubated at 37°C for one hour in the presence of 1 mM 30 methyl- $\beta$ -cyclodextrin (M $\beta$ CD) in order to remove cholesterol from the plasma membranes.

FTIR Spectroscopy and Sample Preparation. Infrared spectra were recorded on a PerkinElmer 2000 Fourier transform IR-spectrometer. Red cell pellets were spread between two CaF<sub>2</sub> infrared windows in a temperature-controlled cell. Intact cells were cooled to -5°C, kept at this temperature for 15 minutes, and then rewarmed to determine the phase transitions. Forty to 50 spectra were recorded over a temperature range from -5°C to +45°C, at a heating rate of 5°C/min. Data processing consisted of taking the second derivative of the IR-absorbance spectra using a 9 point smoothing factor. Inverted second derivative spectra were normalized on the lipid band around 2850 cm<sup>-1</sup>. Band positions were determined by averaging the intercepts at 80% of the absorbance maximum. The wavenumber (cm<sup>-1</sup>) of the CH<sub>2</sub> symmetric stretching vibration was plotted as a function of temperature. The first derivative of the wavenumber versus temperature plots was calculated to show inflections more clearly and are a measure of the cooperativity of the transitions. Phase transition temperatures and cooperativity values were determined from the maxima in the first derivative plots.

FTIR Analysis of Intact Erythrocytes. In situ FTIR analysis of erythrocytes revealed the presence of two clear inflection points in the wavenumber versus temperature plot of the lipid methylene stretching mode at around 2850 cm<sup>-1</sup> (see Figures 24-26). The first derivative of the wavenumber versus temperature plot was calculated to determine the phase transition temperatures. Two transition temperatures were detected in all cases, centered on approximately 14 and 34°C. The thermotropic response depicted in Figure 24 was strikingly different from

that shown in Figures 25 and 26. The total wavenumber excursion in Figure 24 was almost  $1.5\text{ cm}^{-1}$  (from  $-5$  to  $45^{\circ}\text{C}$ ) as compared to  $0.8\text{ cm}^{-1}$  in Figures 25 and 26. The cooperativity of the transitions (maxima in the first derivative plot) in Figure 24 were also strikingly higher compared with the other donors. We discovered that the donor for this experiment used LIPITOR (Pfizer Pharmaceuticals, Ct), and we speculate that this blood cholesterol lowering compound may have affected the erythrocyte membrane properties. Table 1 below presents the average transition temperatures and concomitant cooperativity values. The two transitions in the intact cells at  $14$  and  $34^{\circ}\text{C}$  had cooperativity values of  $0.036$  and  $0.031\text{ cm}^{-1}/^{\circ}\text{C}$ , respectively, suggesting that the two transitions are equally cooperative.

Cholesterol Depletion. M $\beta$ CD was used to remove cholesterol selectively from the erythrocyte membranes. Cholesterol removal had a drastic effect on the thermotropic response of the membrane as is shown in Fig. 27. The wavenumber excursion from  $-5$  to  $45^{\circ}\text{C}$  increased from  $0.7$  to  $2.9\text{ cm}^{-1}$  upon cholesterol removal. At high temperatures, when the membranes are in a disordered phase, the wavenumber of the lipid band increased after cholesterol depletion of the cells and at lower temperatures, when the membranes are in a more ordered phase, cholesterol depletion resulted in a lower wavenumber compared to control cells. This indicates that cholesterol fluidizes the membrane at low temperatures and rigidifies the membrane at high (physiological) temperatures.

The following Table I illustrates the midpoint of phase transitions of M $\beta$ CD-treated and control erythrocytes as derived from FTIR wavenumber (band around  $2850\text{ cm}^{-1}$ ) versus temperature plots:

TABLE I

| Lipids                            | Low Transition | Middle Transition                  | High Transition |
|-----------------------------------|----------------|------------------------------------|-----------------|
|                                   | Transition     | Temperature ( $^{\circ}\text{C}$ ) |                 |
| Control Erythrocytes              | $14.4 \pm 1.3$ |                                    | $34.2 \pm 1.4$  |
| M $\beta$ CD Treated Erythrocytes | $15.3 \pm 0.8$ | $26.0 \pm 0.8$                     | $35.4 \pm 1.5$  |

---

The following Table II illustrates the cooperativity of phase transitions of M $\beta$ CD-treated and control erythrocytes as derived from FTIR wavenumber (band around  $2850\text{ cm}^{-1}$  versus temperature plots:

TABLE II

| Lipids                            | Low Transition    | Middle Transition                  | High Transition   |
|-----------------------------------|-------------------|------------------------------------|-------------------|
|                                   | Cooperativity     | (cm $^{-1}$ / $^{\circ}\text{C}$ ) |                   |
| Control Erythrocytes              | $0.036 \pm 0.013$ |                                    | $0.031 \pm 0.010$ |
| M $\beta$ CD Treated Erythrocytes | $0.051 \pm 0.004$ | $0.095 \pm 0.015$                  | $0.091 \pm 0.020$ |

---

Three phase transitions at about  $15^{\circ}\text{C}$ ,  $26^{\circ}\text{C}$  and  $35^{\circ}\text{C}$  were visible after cholesterol removal from the plasma membranes, and, as expected, the cooperativity of the transitions was greater compared to the control erythrocytic cells. The cooperativity of the transition at about  $15^{\circ}\text{C}$  was found to be only slightly higher than the corresponding transition in the

intact erythrocytes. The cooperativity of the transition at about 35°C, which was also observed in the non-treated control erythrocytic cells, showed a large increase from about 0.031 to about 0.091 cm<sup>-1</sup>/°C after cholesterol removal. The transition at about 26°C was not visible in intact erythrocytes, possibly due to broadening by the abundant cholesterol in the membrane. The cooperativity of this transition was about 0.095 cm<sup>-1</sup>/°C.

#### EXAMPLE 11

Biological membranes may be thought to be in the lamellar liquid-crystalline phase at physiological temperatures. The lipids may be organized in a two dimensional array with acyl chains relatively disordered. At low temperatures, a lamellar gel phase may be formed, in which case lipid acyl chains would be highly ordered and more tightly packed together. A lipid bilayer consisting of one phospholipid species may be characterized by its gel to liquid-crystalline phase transition temperature,  $T_m$ . In a cell membrane, the situation is more complex, because the mixture of lipids, sterols and proteins and the preferential interactions between these components causes a complex thermal phase behavior.

In human erythrocytes there is an asymmetric distribution of phospholipids between both sides of the bilayer membrane. The outer layer contains predominantly phosphatidyl choline and sphingomyelin, whereas the inner layer consists mainly of phosphatidyl ethanolamine and phosphatidyl serine. The lipid composition directly affects membrane fluidity. Other membrane components, such as cholesterol and proteins, also have effects on the membrane. Cholesterol fluidizes the lipid bilayer in the gel state and reduces the motional order in the fluid state.

Moreover, cholesterol preferentially interacts with specific lipids in the membrane, particularly with sphingolipids.

FTIR has been proven to be a very useful method for studying physical properties of membranes both in model systems from isolated lipids as well as *in situ* in whole cells. The wavenumber of the CH<sub>2</sub> stretching mode around 2850 cm<sup>-1</sup> is an indicator of the acyl chain conformational order and may be used to determine phase transitions in cells and tissues. This band mainly arises from endogenous lipids. It has been discovered from FTIR studies on human platelets that these cell fragments have a major phase transition around 15°C and a second transition at around 30°C. Platelets should be stored at 22°C or warmer, well above their main membrane phase transition temperature. In contrast, human erythrocytes may be stored for up to 20 days at 5°C in anticoagulant-preservative solutions.

FTIR may also be used to assess the overall protein secondary structure of intact cells or organisms *in situ*. The application of FTIR to proteins may be based on the assessment of the amide-I band, located between 1600-1700 cm<sup>-1</sup>, and the amide-II band, located between 1600-1500 cm<sup>-1</sup>.

FTIR may further also be used to assess the membrane fluidity and the overall protein secondary structure of erythrocytes *in situ*, thereby omitting the use of interfering probe molecules. As previously indicated methyl-β-cyclodextrin may be used to remove cholesterol selectively from cells and examine the concomitant effect on membrane fluidity. Changes in membrane fluidity and overall protein secondary structure were studied during storage at 4°C. In addition, changes in the membrane lipid composition of the cells were measured using thin layer chromatography, and the formation lipid macro-domains was investigated using fluorescent dye dil-C<sub>18</sub>.



FTIR Analysis of Intact Erythrocytes. Red cells were washed three times in PBS buffer prior to *in situ* IR analyses to reduce interference of plasma components. Figure 28 depicts an enlargement of the IR 3000-2800  $\text{cm}^{-1}$  region of erythrocytes at two different temperatures. The different bands in this region are more clearly visible after taking the second derivative of the absorbance spectra, as best shown in Figure 29. The pronounced bands at 2870 and 2950  $\text{cm}^{-1}$  can be assigned to  $\text{CH}_3$  stretching vibrations of endogenous erythrocyte proteins, substantially hemoglobin. The band at around 2850  $\text{cm}^{-1}$ , which is visible only as a small shoulder next to the protein band (see Figure 28) in the absorbance spectra, is clearly resolved from the protein  $\text{CH}_3$  band in the second derivative spectra, as shown in Figure 29. This band has been assigned to the symmetric methylene stretching vibration of the membrane lipids. The wavenumber of the  $\text{CH}_3$  band did not shift significantly with increasing temperature, whereas the lipid band clearly shifted to a higher wavenumber with increasing temperature, indicating an increase in membrane fluidity (see Figure 30).

Membrane Phase Transitions in Erythrocytes. Two clear inflection points were observed in the wavenumber versus temperature plot of the lipid methylene stretching mode, as best shown in Figure 31 which illustrates the thermotropic response of the symmetric  $\text{CH}_2$  stretch vibration (filled circles) arising from endogenous lipids, and the symmetric  $\text{CH}_3$  stretch vibration (open circles) arising from endogenous proteins in intact erythrocytes. The data were smoothed using a Savitzky-Golay routine, and the first derivative of the wavenumber versus temperature plot was calculated to determine the phase transition temperatures, as best shown in Figure 32 which is a graph of wavenumber versus temperature plots of the  $\text{CH}_2$  symmetric

stretching mode of erythrocytes from one blood donor, along with the first derivatives of the wavenumber versus temperature plots which were employed to determine the phase transition temperatures and concomitant co-operativity values. The thermotropic responses of erythrocytes from two other donors are shown in Figures 33 and 34, to illustrate the variation between three different donors. Figures 33 and 34 illustrate respectively wavenumber versus temperature plots of the  $\text{CH}_2$  symmetric stretching mode of erythrocytes from two additional blood donors, along with first derivatives of the wavenumber versus temperature plots which were used to determine the phase transition temperatures and concomitant co-operativity values. Two transition temperatures were detected in erythrocytes from these three blood donors, centered on approximately 14 and 34°C. The thermotropic response depicted in Figure 32 was different from those shown in Figures 33 and 34. The total wavenumber excursion in Figure 32 was almost  $1.5\text{cm}^{-1}$  (from -5 to 45°C) as compared to  $0.8\text{ cm}^{-1}$  in Figures 33 and 34. The cooperativity of the transitions (maxima in the first derivative plot) in Figure 32 were also higher compared with the two donors shown in Figures 33 and 34. It was discovered that the donor for the experiment in Figure 32 used LIPITOR (Pfizer Pharmaceuticals, CT), and it is believed that this blood cholesterol lowering compound may have affected the erythrocyte membrane properties. The thermotropic responses depicted in Figures 33 and 34 are more typical for erythrocytes.

The following Table III broadly presents the average transition temperatures and concomitant cooperativity values of five different donors.

TABLE III

| Lipids                            | Low Transition  | Middle Transition                  | High Transition   |
|-----------------------------------|---|------------------------------------|-------------------|
|                                   | Transition  | Temperature ( $^{\circ}\text{C}$ ) |                   |
| Control Erythrocytes              | 14.4 $\pm$ 1.3  |                                    | 34.2 $\pm$ 1.4    |
| M $\beta$ CD Treated Erythrocytes | 15.3 $\pm$ 0.8  | 26.0 $\pm$ 0.8                     | 35.4 $\pm$ 1.5    |
|                                   | Cooperativity ( $\text{cm}^{-1} / ^{\circ}\text{C}$ ) |                                    |                   |
| Control Erythrocytes              | 0.036 $\pm$ 0.013                                     |                                    | 0.031 $\pm$ 0.010 |
| M $\beta$ CD Treated Erythrocytes | 0.051 $\pm$ 0.004                                     | 0.095 $\pm$ 0.015                  | 0.091 $\pm$ 0.020 |

---

The foregoing Table III more specifically illustrates cooperativity and midpoint of phase transitions of M $\beta$ CD-treated and control erythrocytes as derived from FTIR wavenumber (band around 2850  $\text{cm}^{-1}$  versus temperature plots, with their associated standard error, for five blood donors. The two transitions (i.e., the low and high transitions) in the intact cells at 14 and 34 $^{\circ}\text{C}$ , had cooperativity values of 0.036 and 0.031  $\text{cm}^{-1}/^{\circ}\text{C}$ , respectively, suggesting that the two transitions are equally cooperative.

Cholesterol Depletion. M $\beta$ CD was used to remove cholesterol from the erythrocyte membranes. Cholesterol removal had a drastic effect on the thermotropic response of the membrane, as best shown in Figure 35. The wavenumber excursion from -5 to 45 $^{\circ}\text{C}$  increased from 0.7 to 2.9  $\text{cm}^{-1}$  upon cholesterol removal. At high temperatures, when the membranes are in a disordered state, the wavenumber of the lipid band increased after cholesterol depletion of the cells and at lower temperatures, when the

membranes are in a more ordered state, cholesterol depletion resulted in a lower wavenumber compared to control cells. This indicates that cholesterol fluidizes the membrane at low temperatures and rigidifies the membrane at high (physiological) temperatures.

The three transitions at 15, 26, and 35°C were visible after cholesterol removal from the plasma membranes, and, as expected, the cooperativity of the transitions was greater compared to the control cells. The cooperativity of the transitions was greater compared to the control cells. The cooperativity of the transition at 15°C was found to be slightly higher than the corresponding transition in the intact erythrocytes. The cooperativity of the transition at 35°C, which was also observed in the non-treated control cells, showed a large increase from 0.031 to 0.091  $\text{cm}^{-1}/^{\circ}\text{C}$  after cholesterol removal. The transition at 26°C was not visible in intact erythrocytes, possibly due to broadening by the abundant cholesterol in the membrane. The cooperativity of this transition was 0.095  $\text{cm}^{-1}/^{\circ}\text{C}$ .

Ghosts. The thermotropic response of erythrocyte ghosts were studied to corroborate the results with the intact erythrocytes. As best shown in Figure 36, the wavenumber of the  $\text{CH}_2$  stretching band in ghosts gradually increased from 2851.7 to 2852.6  $\text{cm}^{-1}$  when the temperature was increased from -5 to 45°C. Four minor transitions at approximately 3, 15, 24 and 36°C were visible after first derivative analysis. The cooperativity of these transitions was lower compared with those in the intact erythrocytes. Again, cholesterol removal enhanced the cooperativity of the transitions, which were observed at slightly different temperatures, at 4, 12, 20 and 30°C, compared with the non-M $\beta$ CD treated ghosts. The differences in

thermotropic response of ghosts compared with the intact erythrocytes suggest a rearrangement of the membrane lipids (and proteins) upon hemolysis.

Effect of Storage on Thermotropic Response. Erythrocytes were stored in autologous acid citrate dextrose (ACD) anticoagulated platelet poor plasma at 4°C. Aliquots of the cells were washed three times in cold PBS buffer, and the effect of cold storage on the membrane fluidity was studied, using FTIR. The wavenumber of the lipid band at 4°C dropped significantly after 5 days of cold storage, as best shown in Figure 37, suggesting a rigidification of the membrane. In addition, the thermotropic response of the cells also changed during cold storage (see Figure 38). After one day, the two main transitions shifted to lower temperatures, and after five days storage, multiple sharp phase transitions were observed, indicating a large scale rearrangement of the membrane lipids.

Formation of Lipid Domains. Lipid dye dil-C<sub>18</sub> was used to study if regrouping of lipids during cold storage as suggested by the FTIR experiments, was accompanied by formation large membrane domains or aggregates. Dil-C<sub>18</sub> preferentially partitions into ordered lipid domains and can be used to investigate lipid phase separation in living cells. Figure 39 depicts an image of dil-C<sub>18</sub> labeled erythrocytes after 4 days exposure to 4°C. The dye remained uniformly distributed during cold storage (only the results after 4 days are shown) which suggests that the lipid domains in the membrane are below the resolution limit of the microscope.

Changes in Lipid Compositions. The composition of diacyl phospholipids and the amount of cholesterol in the total lipid extract of human erythrocytes is presented in the following Table IV:

TABLE IV

| Storage<br>Time   | FFA     | Chol.    | PE      | PS      | PC       | SM       |
|-------------------|---------|----------|---------|---------|----------|----------|
| Fresh             |         | 48.1±2.2 | 6.2±2.7 | 2.6±0.2 | 24.2±0.4 | 18.9±0.3 |
| 1 day<br>storage  | 5.0±0.5 | 36.7±0.1 | 7.2±0.2 | 5.1±0.2 | 25.8±0.0 | 20.2±0.0 |
| 5 days<br>storage | 2.4±0.2 | 43.6±1.3 | 6.4±0.2 | 4.2±0.2 | 24.2±0.5 | 18.8±0.3 |

The foregoing Table IV more specifically presents lipid composition of human erythrocytes, with the values representing the mean of two samples express as weight percent of the total lipids and with the abbreviations having the following meaning: Chol, cholesterol; FFA, free fatty acids; PE, phosphatidylethanolamine; PS, phosphatidylserine; PC, phosphatidylcholine; SM, shingomyelin. As shown in Table IV, phosphatidylcholine is the most abundant lipid (24 wt.%). The amounts of PE and PS are 6 and 3 wt.% respectively. Sphingolipids make up 19 wt.% of the lipid composition. The cholesterol content was found to be approximately 50 wt.% in fresh erythrocytes. Even after one day storage, free fatty acids were detected in the erythrocyte membrane, indicating chemical changes in the membrane composition. In addition, the lipid analysis revealed that the amount of cholesterol decreased during *in vitro* storage at 4°C. The relative amounts of the phospholipids did not change significantly during storage.

Overall Protein Secondary Structure. The overall protein secondary structure of erythrocytes during cold storage was monitored using the IR amide-II band. FTIR measures an average signal of all cellular proteins, but in erythrocytes the amide-

II band is dominated by the hemoglobin. The shape of the amide-II band with a major band at  $1550\text{ cm}^{-1}$ , and only minor contributions of bands at  $1535\text{ cm}^{-1}$  and  $1565\text{ cm}^{-1}$  suggests a high relative proportion of  $\alpha$ -helical content of hemoglobin. No changes in amide-II band profile were observed after 9 days of storage at  $4^{\circ}\text{C}$ .

### Discussion

Physical properties of membranes from human erythrocytes were evaluated using FTIR, and it has been demonstrated that substantial changes in the membrane physical properties and lipid composition occur during cold storage. In contrast, the average protein secondary structure of erythrocyte hemoglobin was not affected by cold storage.

The human erythrocyte was the first cell in which it was shown that phospholipids are organized in domains within the membrane rather than being homogeneously distributed. The asymmetric distribution with predominantly PS and PE in the inner leaflet and, PC and SM in the outer leaflet was demonstrated using specific phospholipids in the erythrocyte membrane with a more rigid outer leaflet and a more fluid inner leaflet.

Cholesterol modulates lipid intermixing in lipid bilayers: phospholipids which are strongly phase separated in the absence of cholesterol become homogeneously mixed at high (e.g., 50 mol%) cholesterol contents. Therefore, the observation of multiple transitions in erythrocytes is up-expected. However, the specific mixture of sphingolipids and phospholipids in the erythrocyte membrane may explain why multiple phase transitions can be observed at such high cholesterol contents. Mixtures of phospholipids and glycosphingolipids exhibit significant

inhomogeneity in lipid mixing even at high (50%) bilayer cholesterol contents.

Cholesterol depletion confirmed that the small inflections that were observed in the thermal profiles of intact cells correspond to real thermal events in the membrane. Three major cooperative transitions became visible in the wavenumber versus temperature plot after cholesterol depletion of erythrocyte membranes.

It is suggested that the transition at 34°C observed both in intact and in cholesterol depleted cells reflects the melting of the more rigid sphingolipid rich outer leaflet of the membrane because sphingolipids generally exhibit relatively high melting points, ranging from about 25°C to 45°C. Sheep erythrocyte membranes exhibit a sphingolipid transition between 26°C and 35°C. It is suggested that the transition at 15°C reflects the melting of the more fluid inner membrane leaflet. The phase transitions of red cells which were observed are very similar to those of human platelets, which have two transitions at around 15°C and 30°C.

The transition at 26°C in the cholesterol depleted cells was most likely masked by the 30 cholesterol in the intact cells, or alternatively, it may reflect a mixed phase due to regrouping of the membrane lipids between the inner and outer membrane leaflet. The data support the concept that cholesterol modulates the membrane fluidity by decreasing the lipid order at low temperatures and by increasing the lipid order at high temperatures.

Thus, the FTIR studies reveal that red blood cells and platelets have similar phase transitions and in both cell types phase separation was observed upon long-term exposure to cold



temperatures. But in contrast to platelets, no large membrane aggregates were formed in the erythrocyte membrane during cold storage. It is suggested that the low level of aggregation of microdomains in red blood cells may be responsible for their relative insensitivity to chilling damage. That property could result from the high proportion of cholesterol in red blood cell membranes compared with that seen in platelets.

#### EXAMPLE 12

Isolation of Erythrocytes, Cholesterol, Depletion, Preparation of Ghosts, and Storage Conditions: Venous blood was collected from healthy adults, with informed consent, according to institutional protocols. Blood was anticoagulated with ACD (citric acid, citrate, dextrose). Whole blood was centrifuged at  $320 \times g$  for 8 minutes. Platelets were isolated from the platelet rich plasma and used for other experiments. Platelet poor plasma was added back to the erythrocytes and the mixture was stored at  $4^{\circ}\text{C}$ .

Cells were incubated at  $37^{\circ}\text{C}$  for one hour in the presence of 10 mM methyl- $\beta$ -cyclodextrin ( $\text{M}\beta\text{CD}$ ) in order to remove cholesterol from the plasma membranes. Erythrocyte ghosts were prepared. ACD anticoagulated blood was stored in 15 ml polypropylene Falcon tubes at  $4^{\circ}\text{C}$ . Aliquots were taken, and the cells were washed at least three times in cold PBS buffer (100 mM NaCl, 9.4 mM  $\text{Na}_2\text{HPO}_4$ , 0.6 mM  $\text{KH}_2\text{PO}_4$ , pH 7.4), prior to further analyses.

Lipid Extraction, Separation and Analysis. Erythrocyte lipids were extracted in methanol/chloroform (1:2, v/v), and incubated on ice for 1 hour. The lipid composition of erythrocytes was analyzed by rod-typed analytical thin layer

chromatography using a Latroscan TH-10 from (Lantron Laboratories of Tokyo, Japan). The traces were analyzed using Sigmaplot from Jandel Scientific of San Rafael, CA. The individual lipids were expressed as weight per cent of the total lipids.

FTIR Spectroscopy and Sample Preparation. Infrared spectra were recorded on a Perkin Elmer 2000 Fourier transform IR-spectrometer as described previously. Red cell pellets were spread between two  $\text{CaF}_2$  infrared windows in a temperature-controlled cell. Intact cells were cooled to  $-5^\circ\text{C}$ , kept at this temperature for 15 minutes, and then rewarmed to determine the phase transitions. Forty to 50 spectra were recorded over a temperature range from  $-5^\circ\text{C}$  to  $+45^\circ\text{C}$ , at a heating rate of  $5^\circ\text{C}/\text{min}$ . Four scans were accumulated for each spectrum between  $3600\text{--}900\text{ cm}^{-1}$  at  $4\text{ cm}^{-1}$  resolution.

Data processing consisted of taking the second derivative of the IR-absorbance spectra using a 9 point smoothing factor. Inverted second derivative spectra were normalized on the lipid band around  $2850\text{ cm}^{-1}$ . Band positions were determined as described previously. The wavenumber ( $\text{cm}^{-1}$ ) of the  $\text{CH}_2$  symmetric stretching vibration was plotted as a function of temperature. The first derivative of the wavenumber *versus* temperature plots was obtained using Peakfit from Jandel Scientific, San Rafael, CA to show inflections more clearly and as a measure of the cooperativity of the transitions. Phase transition temperatures and cooperativity values were determined from the maxima in the first derivative plots. For the protein studies, the spectral region between  $1600$  and  $1500\text{ cm}^{-1}$  selected. This region contains the amide-II absorption band of the protein backbones.

Fluorescence Microscopy. Washed erythrocytes were diluted to a final concentration of  $1 \times 10^8$  cells/ml. They were labeled

with dil-C<sub>18</sub> from Molecular Probes, Inc. of Eugene, OR (2.5  $\mu$ /ml) for 60 minutes in the dark at 37°C. Aliquots were incubated at 37°C or stored at 4°C. The labeled erythrocytes were then fixed with 1 % paraformaldehyde for two hours at the corresponding temperatures. The cells were placed on microscope slides and examined with a Zeiss ICM405 inverted microscope (Planachromat 100X/ 1.4 n.a.objective) and photographed with Ektachrome 400 film from Kodak, Rochester, NY.

### Conclusion

Embodiments of the present invention provide that trehalose, a sugar found at high concentrations in organisms that normally survive dehydration, can be used to preserve biological structures in the dry state. Human blood platelets can be loaded with trehalose under specified conditions, and the loaded cells can be freeze dried with excellent recovery. Additional embodiments of the present invention provide that trehalose may be used to preserve nucleated (eukaryotic) cells.

Eukaryotic cells lines, such as human mesenchymal stem cells and a epithelial 293H cells, have two membrane phase transitions at approximately 15°C and 35°C. Further, they are able to take up solutes from an extracellular medium, as indicated by their loading with the fluorescent dye Lucifer yellow CH. This technique may be employed to load cells with an oligosaccharide, preferably trehalose. Trehalose does not interfere with the growth and viability of cells for up to three days. Cells loaded with trehalose and freeze-dried were viable immediately following rehydration and were healthy in that the membranes appeared intact and the nuclei were clearly visible and were of normal morphology. Some cells even attached weakly

to the substrate and appeared in relatively good physical shape even after 5 days post-rehydration.

Alcohol-reduced erthrocytes have three membrane phase transitions at approximately 15°C, 26°C and 35°C. Alcohol (e.g. cholesterol) depletion of erthrocytes resulted in a large increase in the cooperativity of the membrane phase transitions. Any of the membrane phase transitions, especially the phase transitions at around 35°C, may be used to load erthrocytes with a protectant (e.g. an oligosaccharide such as trehalose). Alcohol-reduced erthrocytes loaded with a protectant and freeze-dried were viable immediately following rehydration.

While the present invention has been described herein with reference to particular embodiments thereof, a latitude of modification, various changes and substitutions are intended in the foregoing disclosure, and it will be appreciated that in some instances some features of the invention will be employed without a corresponding use of other features without departing from the scope and spirit of the invention as set forth. Therefore, many modifications may be made to adapt a particular situation or material to the teachings of the invention without departing from the essential scope and spirit of the present invention. It is intended that the invention not be limited to the particular embodiment disclosed as the best mode contemplated for carrying out this invention, but that the invention will include all embodiments and equivalents falling within the scope of the appended claims.